

# Ammonia and its influence on nitrogen deposition and fine particle formation

Jeffrey L. Collett, Jr.

Atmospheric Science Department, Colorado State University

# Acknowledgments

- People

- CSU: Katie Benedict, Yi, Li, Tony Prenni, Ashley Evanoski-Cole, Derek Day, Florian Schwandner, Suresh Raja, Taehyoung Lee, Amy Sullivan, Kip Carrico, Sonia Kreidenweis
- NPS: Bret Schichtel, William Malm
- ARS: Mark Tigges, John Molenaar, Stephen Holcomb, Cassie Archuleta, Lincoln Sherman
- Shell: Angela Zivkovich
- UC Davis: Chuck McDade, Jose Mojica, Mark van de Water
- RTI: Eva Hardison, David Hardison

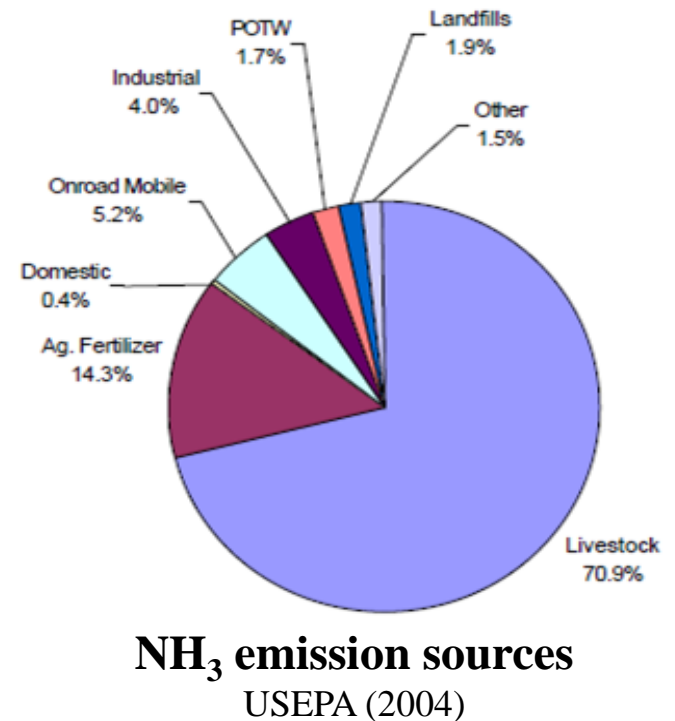
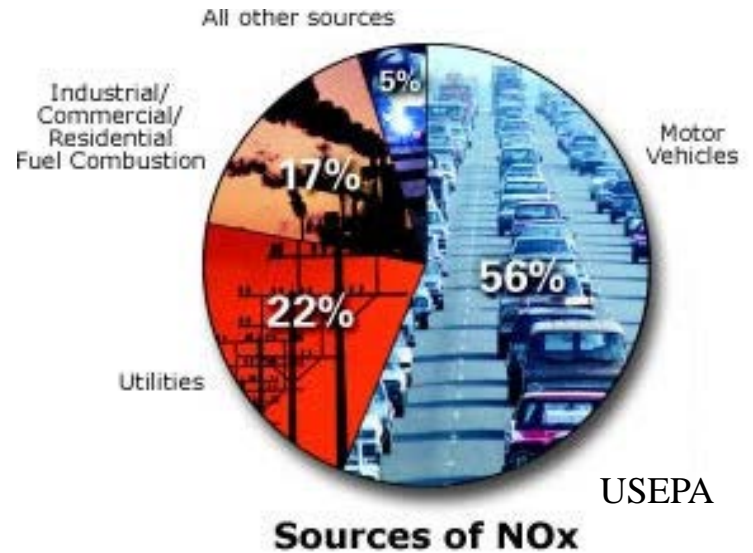
- Funding

- U.S. National Park Service
- U.S. Department of Agriculture
- Shell Exploration and Production Company



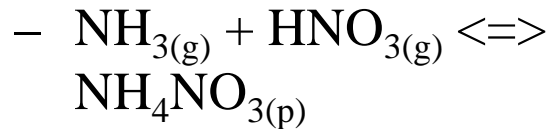
# Reactive nitrogen emissions

- Nitrogen oxides
  - $\text{NO}_x = \text{NO} + \text{NO}_2$
  - Formed by high temperature reaction of  $\text{N}_2$  and  $\text{O}_2$
  - $\text{NO}_x$  reacts in the atmosphere to form nitric acid ( $\text{HNO}_3$ ) and other species
- Ammonia ( $\text{NH}_3$ )
  - Livestock and fertilizer are largest sources



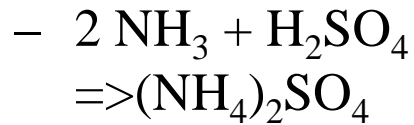
# Particulate atmospheric nitrogen

- Ammonium nitrate



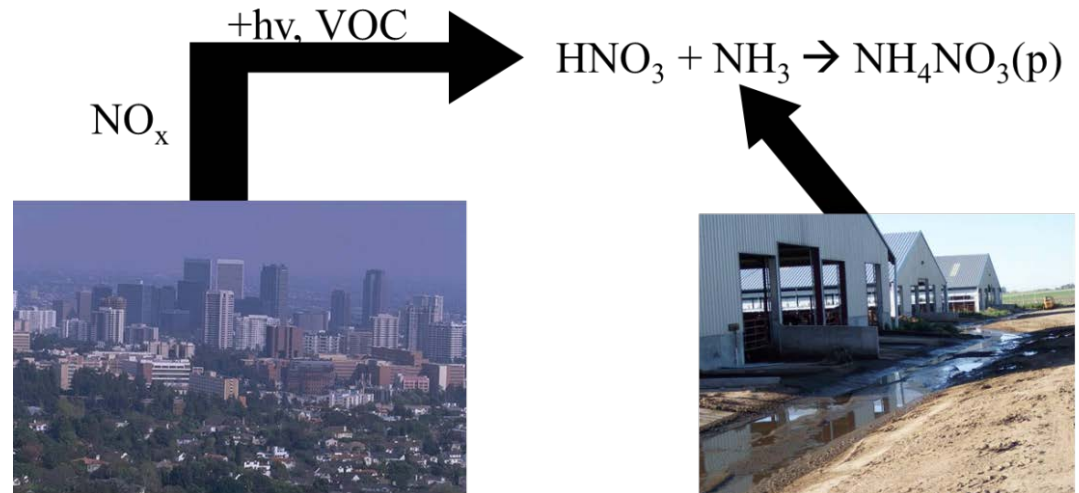
- Sensitive to temperature and relative humidity

- Ammonium sulfate



- Particles ~200-600 nm in diameter

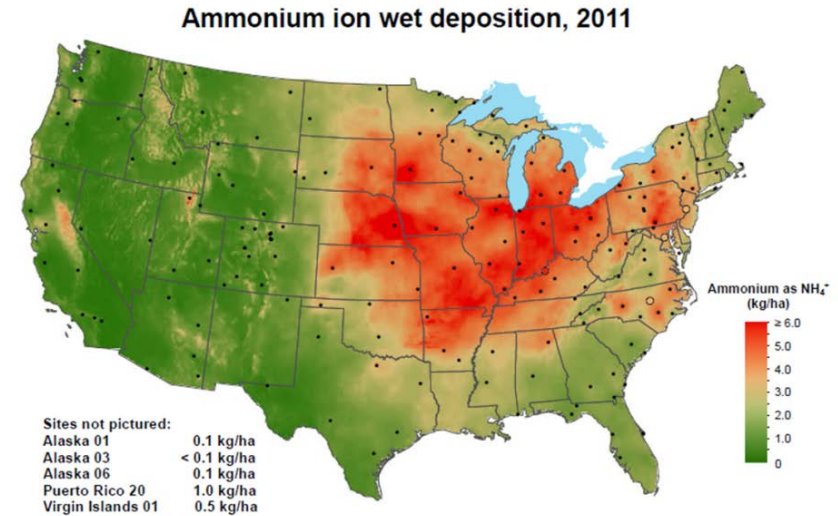
- Long lifetimes in atmosphere (several days)
  - Important cause of haze



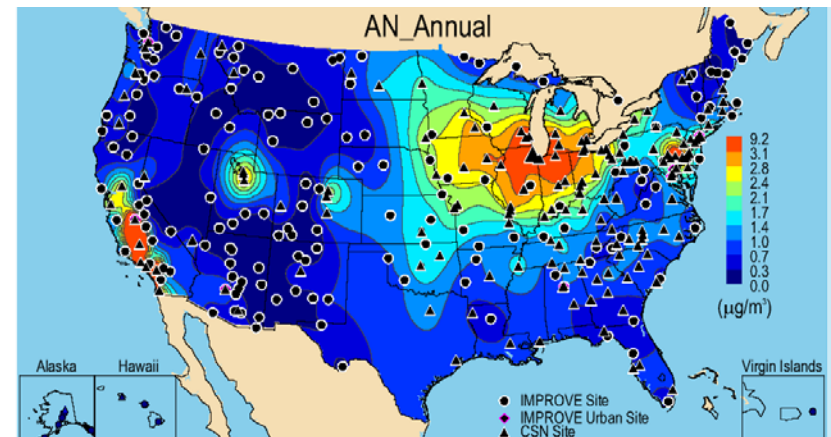
Denver brown cloud (Denver Post)

# Reactive nitrogen deposition

- Wet deposition
  - Precipitation scavenges gases and particles and deposits them on surface
  - Easier to measure
- Dry deposition
  - Difficult/expensive to measure
  - Estimate as product of measured concentration and modeled deposition velocity
  - Gas/particle partitioning is key
    - $V_{d,NH_3} \gg V_{d,NH_4^+}$
    - $V_{d,HNO_3} \gg V_{d,NO_3^-}$



National Atmospheric Deposition Program/National Trends Network  
<http://nadp.isws.illinois.edu>

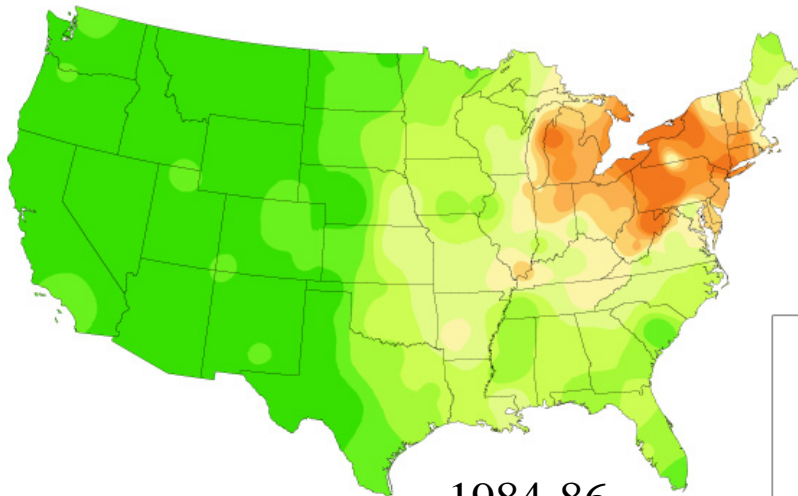


Hand et al. (2011) IMPROVE report

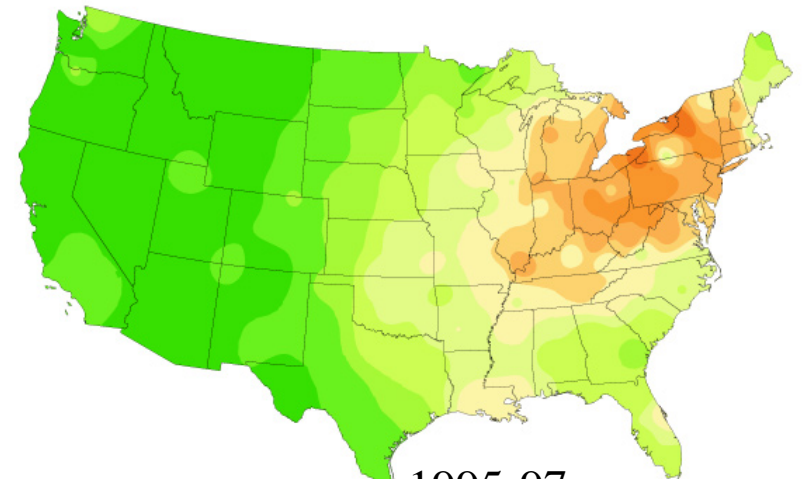
# **HISTORICAL CHANGES IN REACTIVE N WET DEPOSITION**



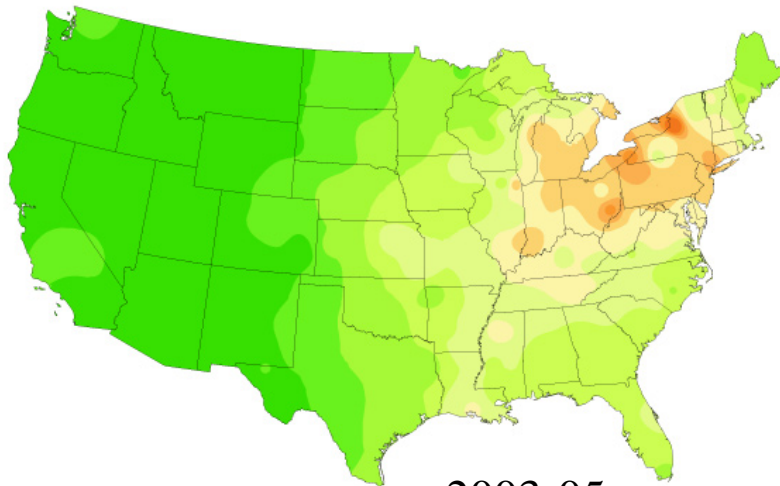
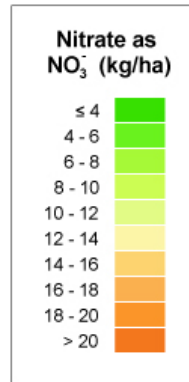
# Nitrate Ion Wet Deposition



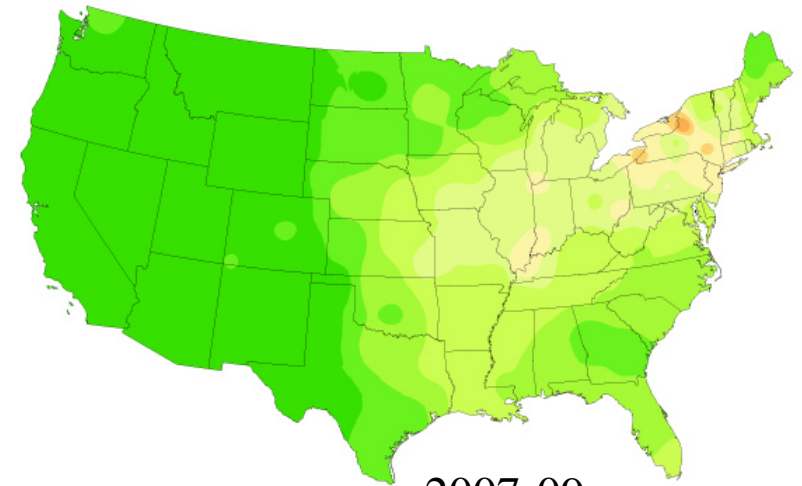
1984-86



1995-97

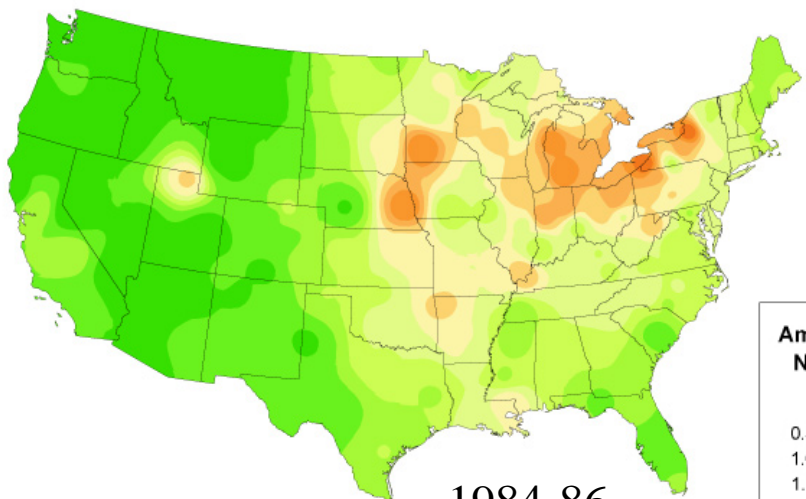


2003-05

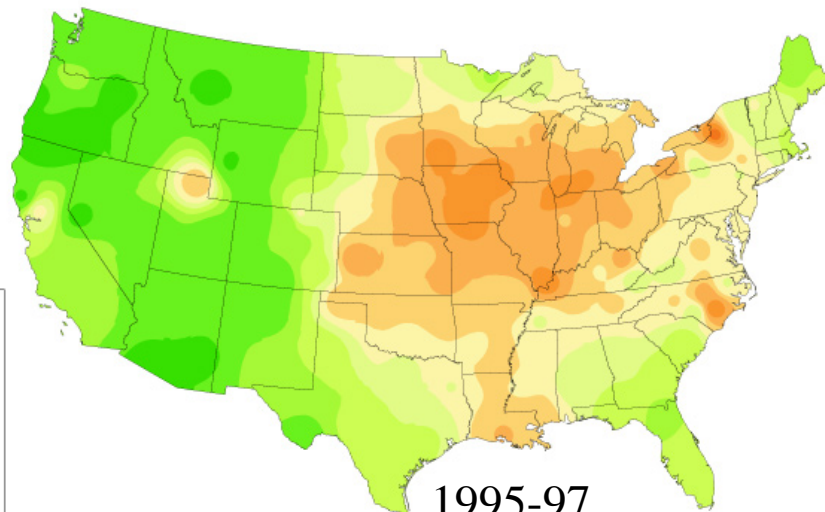


2007-09

# Ammonium Ion Wet Deposition

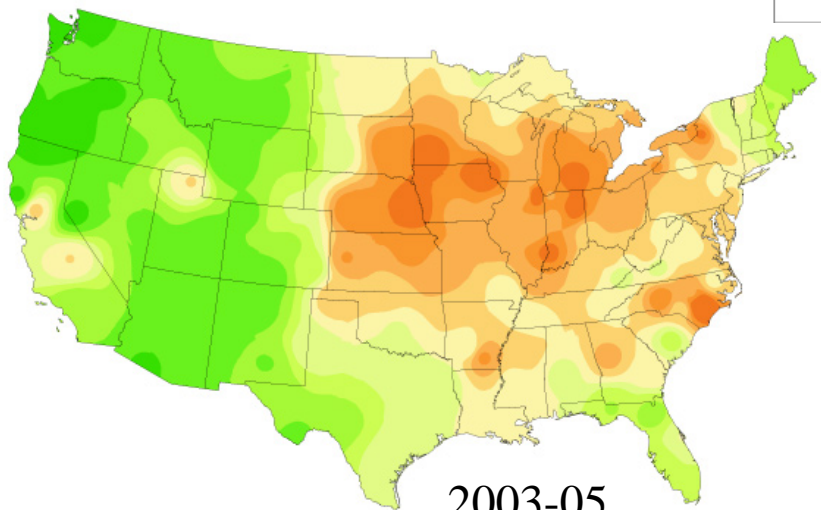
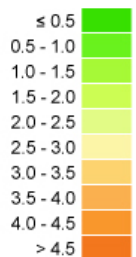


1984-86

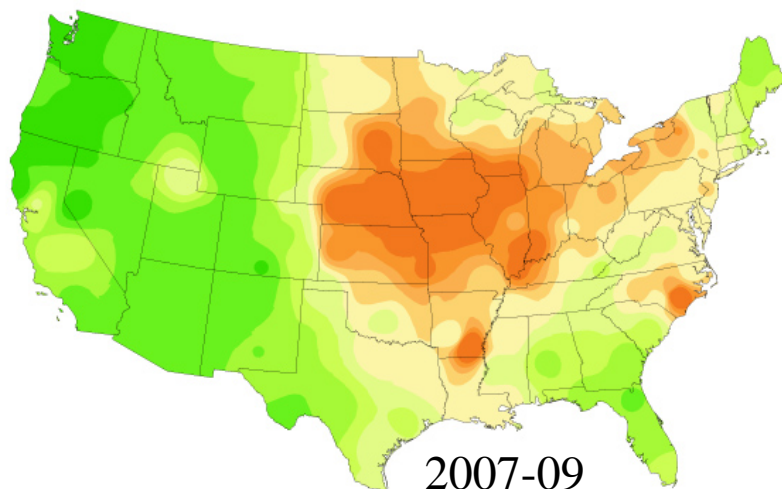


1995-97

Ammonium as  
 $\text{NH}_4^+$  (kg/ha)



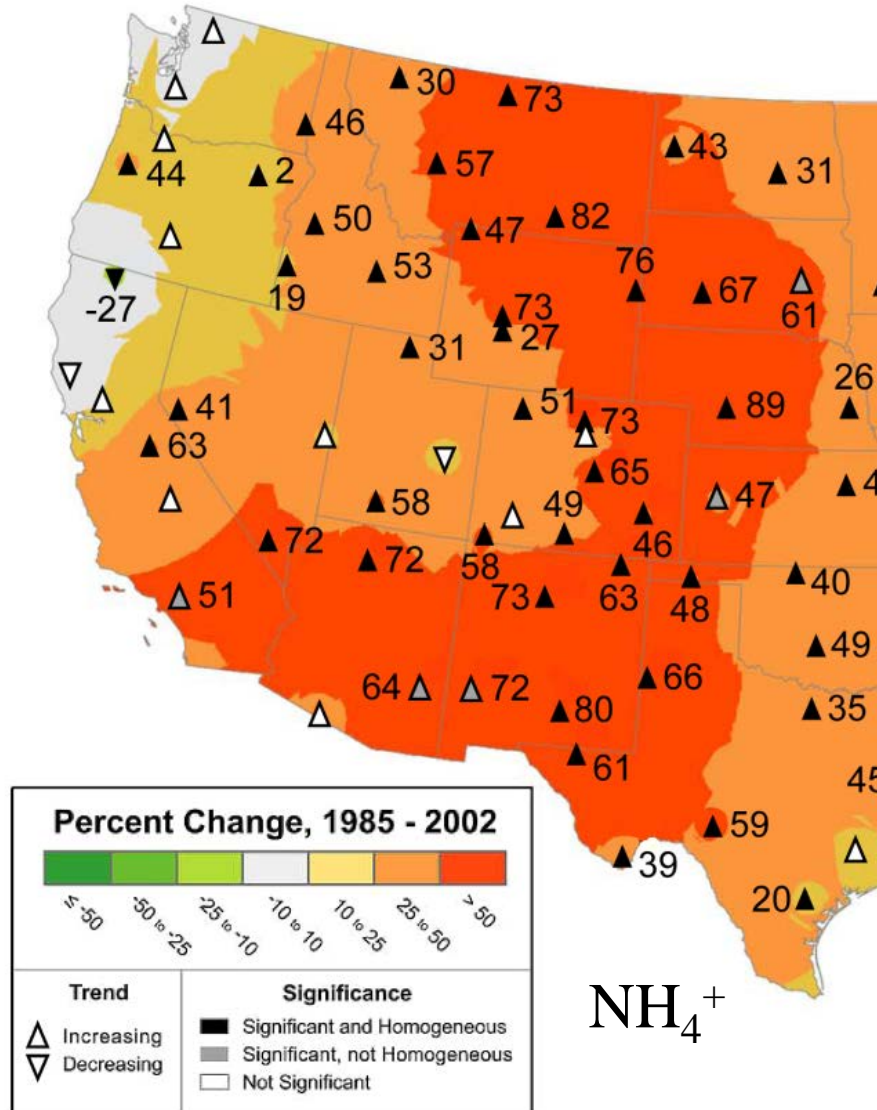
2003-05



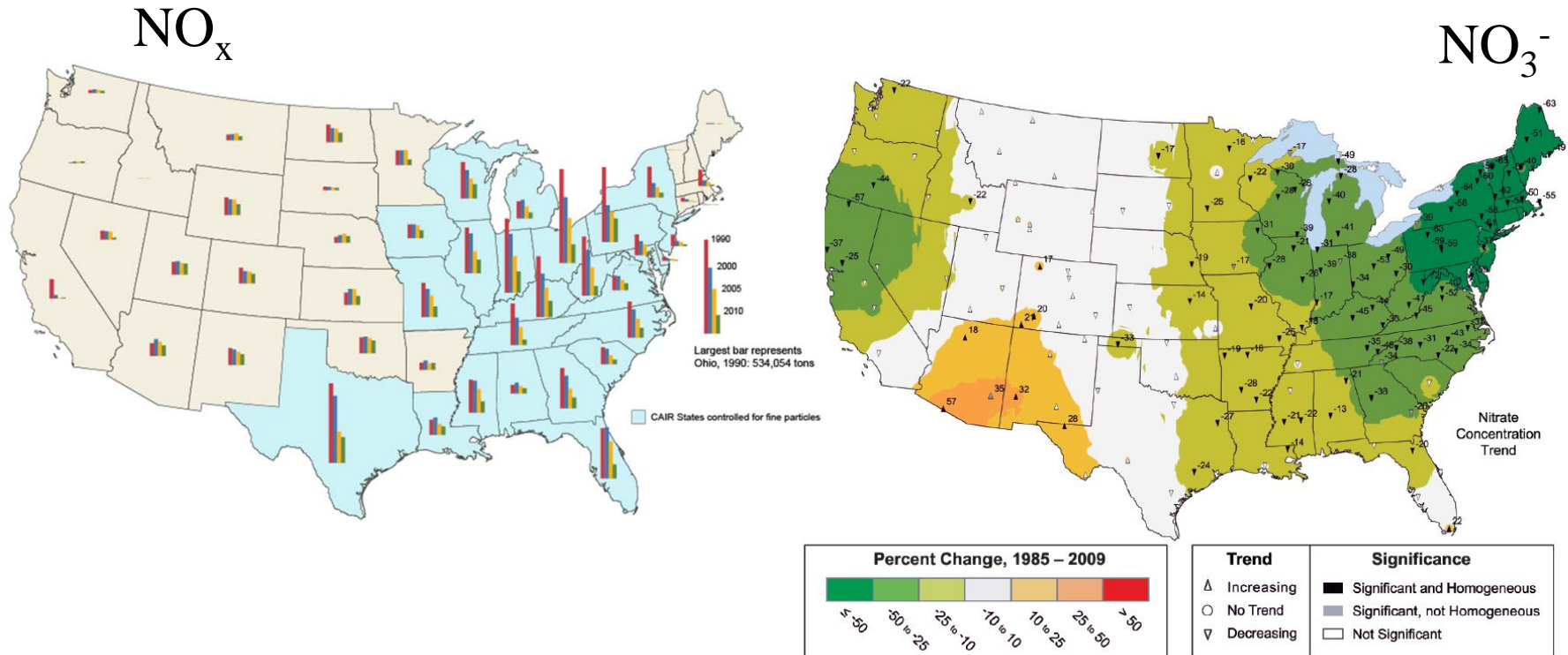
2007-09



# Changes in $\text{NH}_4^+$ wet deposition

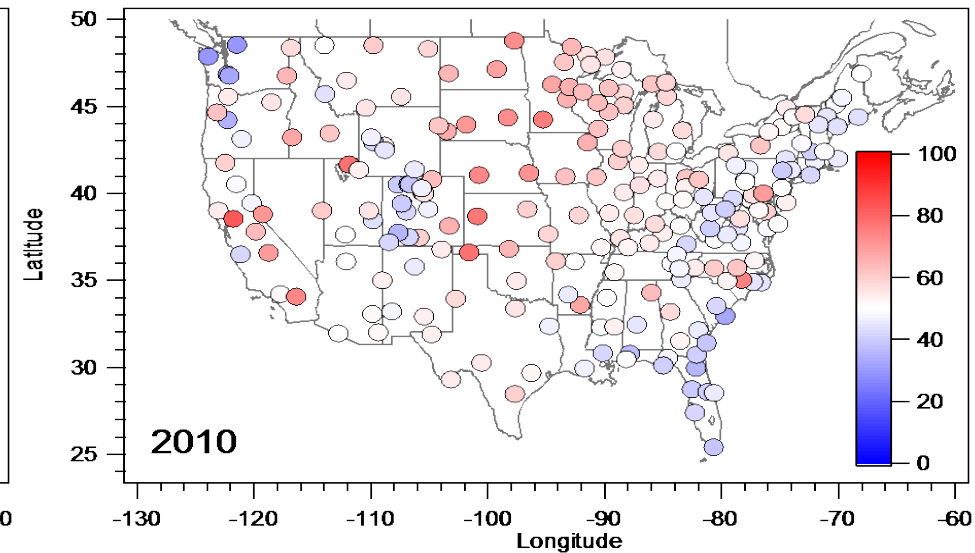
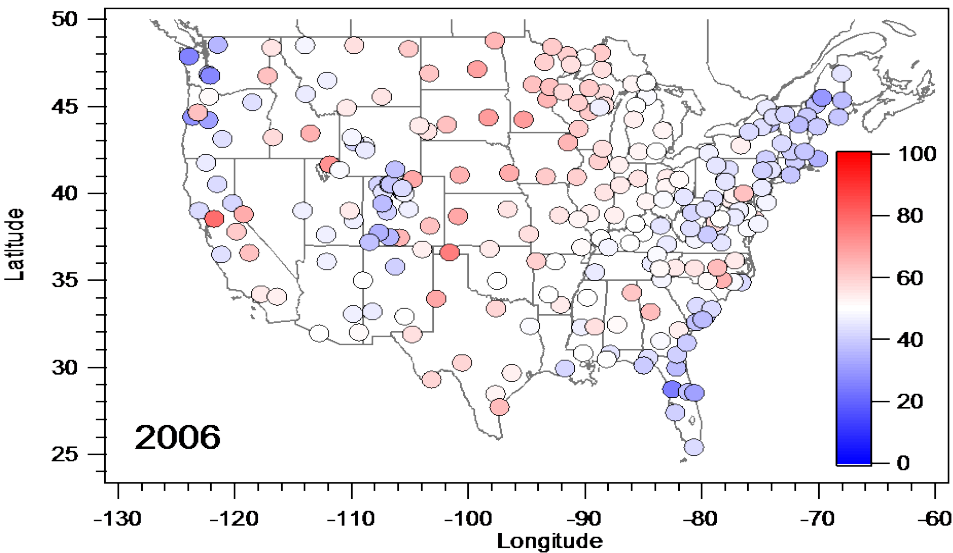
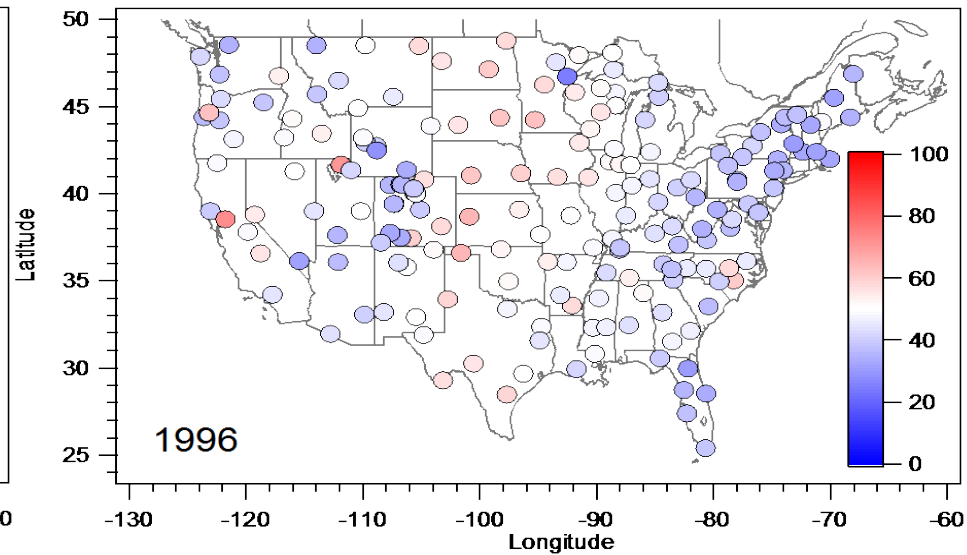
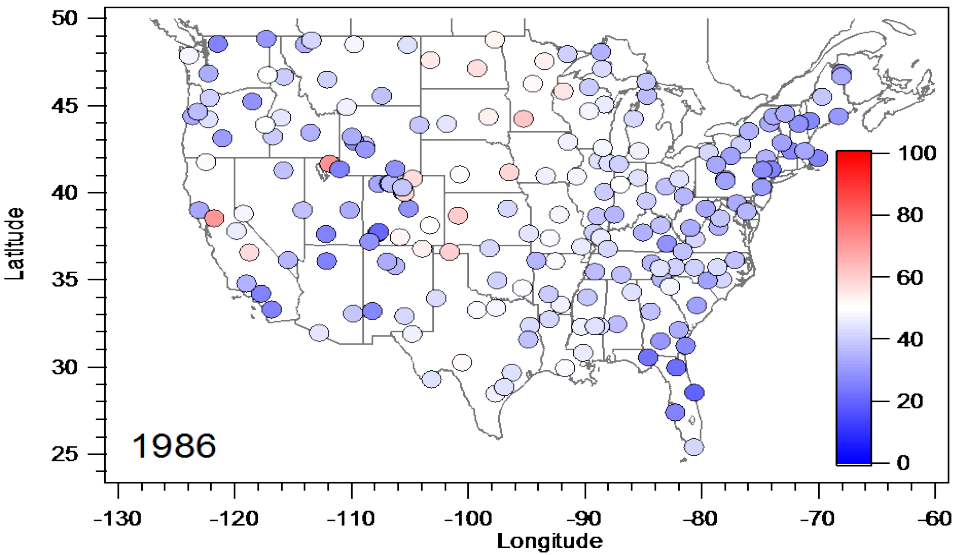


# Changes in $\text{NO}_3^-$ wet deposition and $\text{NO}_x$ emissions



Lehmann and Gay, 2011

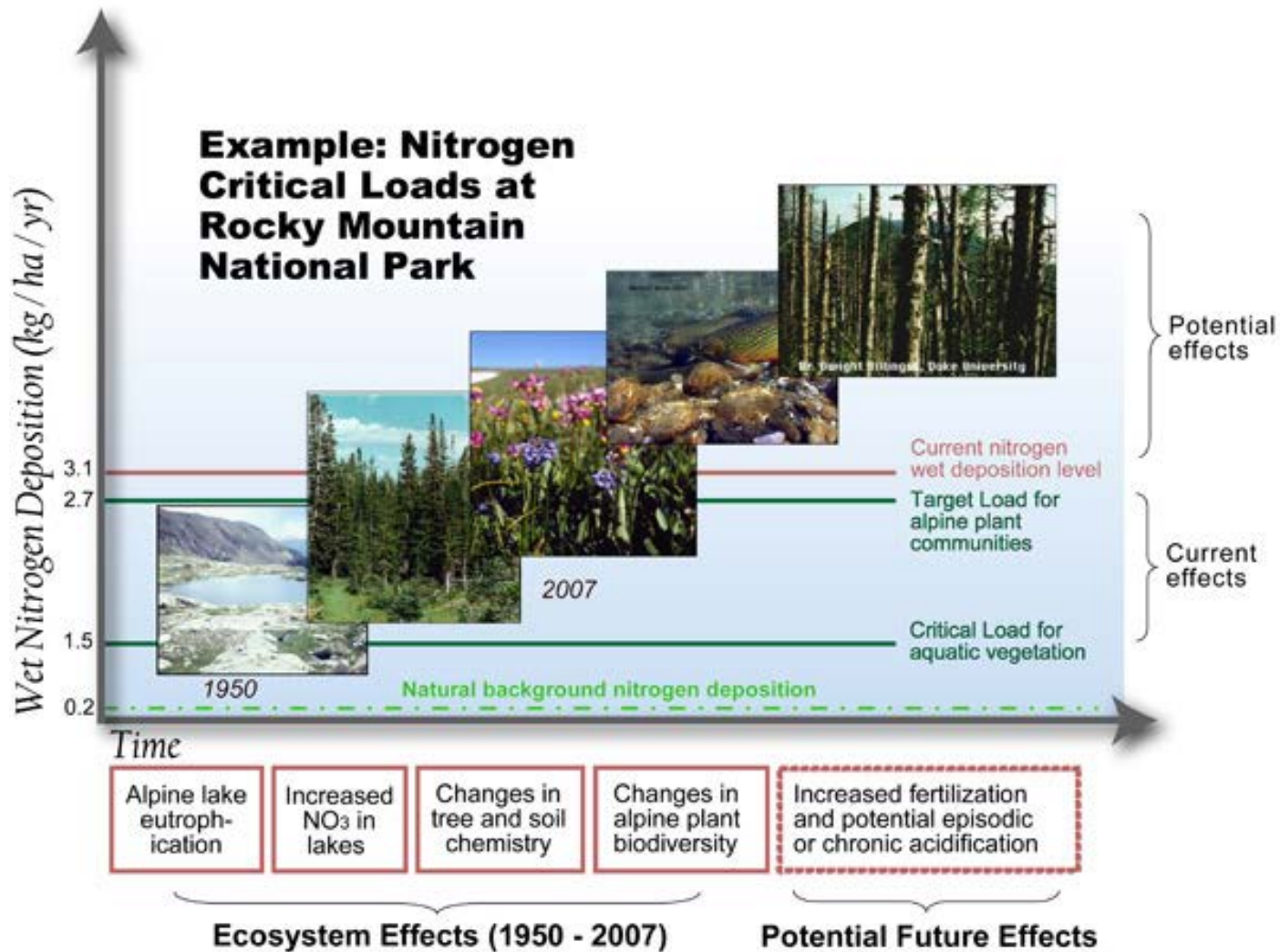
# Change in $\text{NH}_4^+$ fraction in U.S. wet inorganic N deposition



$$\text{NH}_4\% = \frac{\text{NH}_4^+}{\text{NO}_3^- + \text{NH}_4^+} \times 100\%$$

**THE IMPORTANCE OF NH<sub>3</sub>  
FOR N DEPOSITION AND  
PARTICLE FORMATION: A  
FEW EXAMPLES**

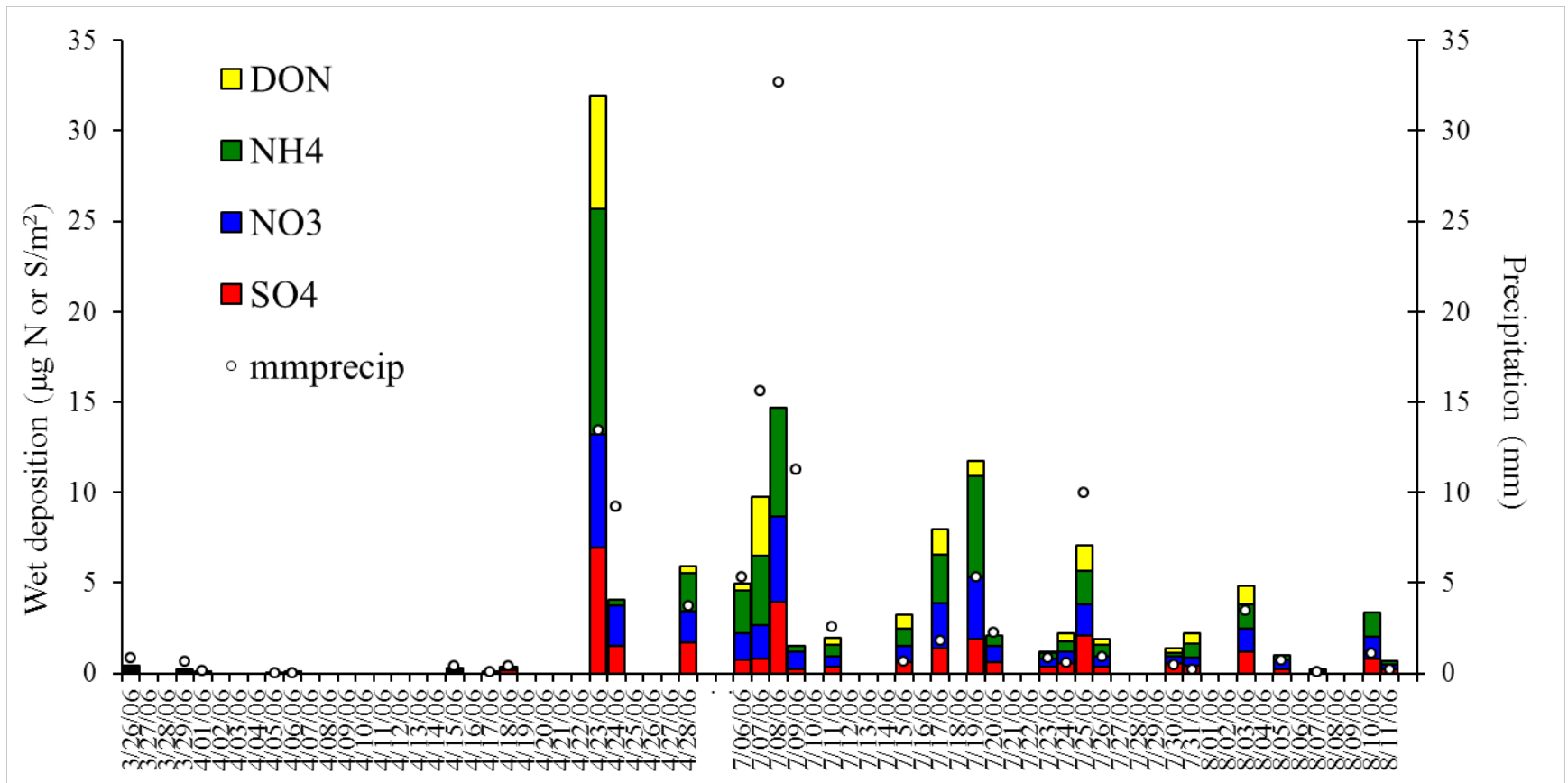
# Concerns about nitrogen deposition







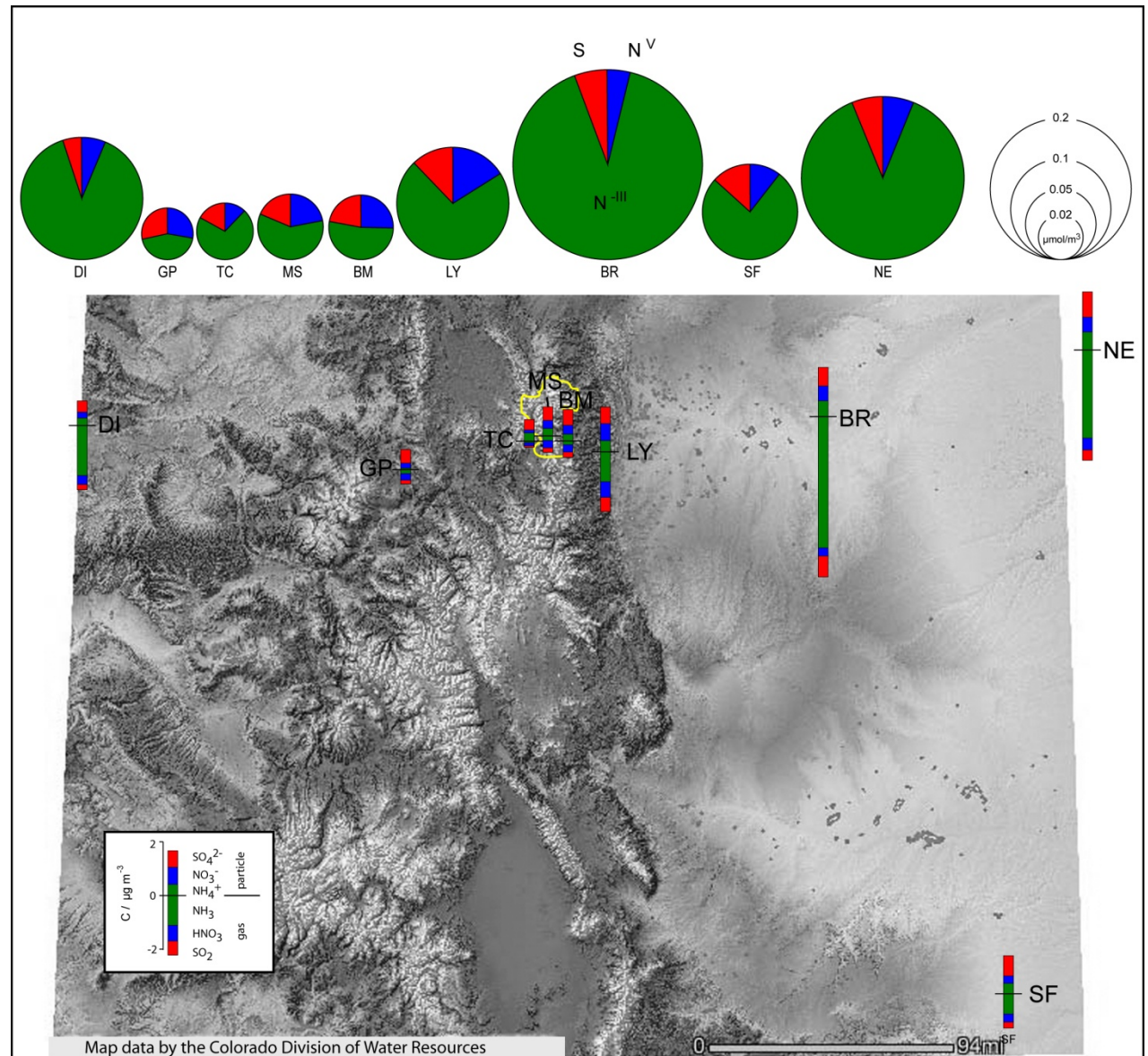
# RoMANS 2006 core site wet deposition



- Spring flux dominated by single event -- summer flux contributed by several events
- Substantial oxidized, reduced, and organic N

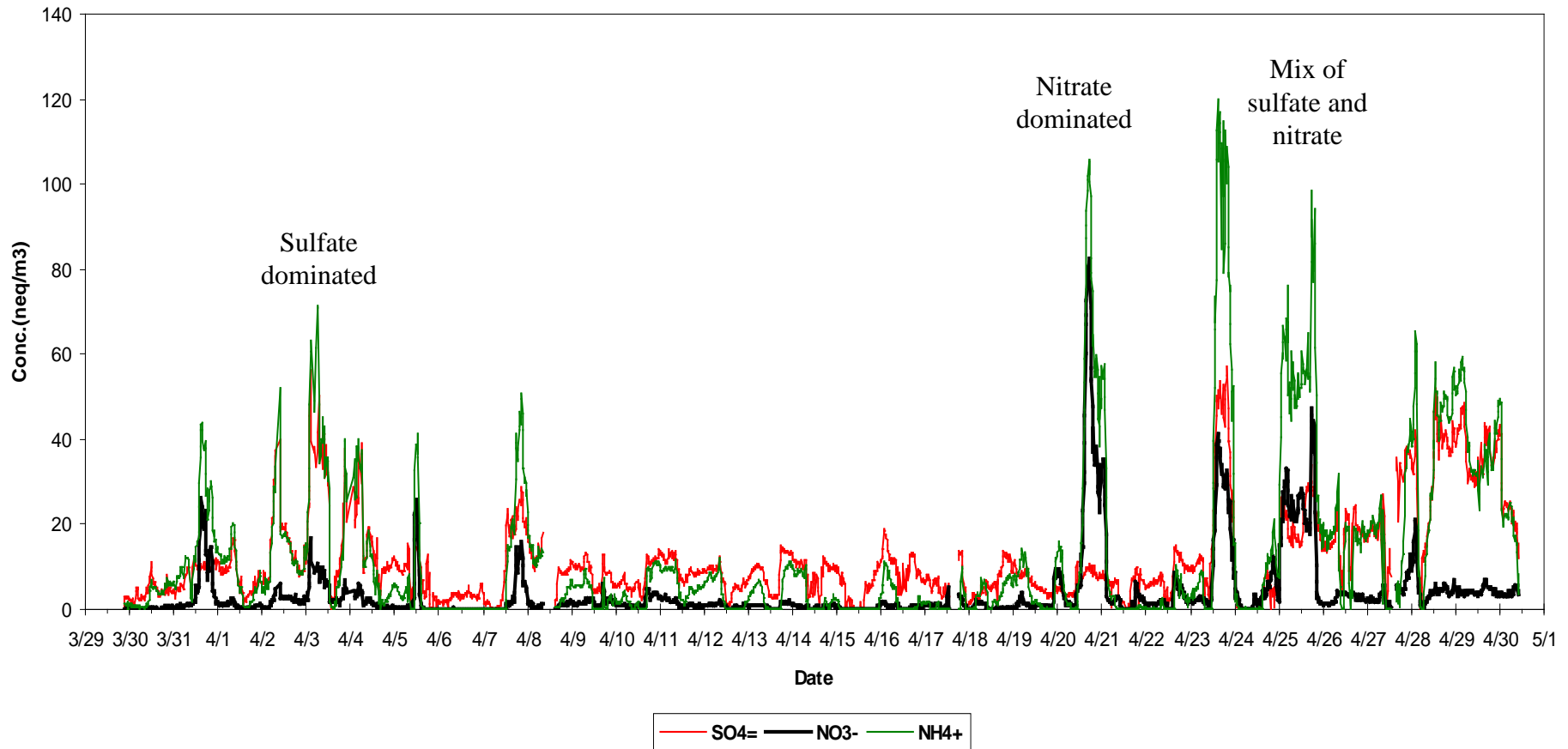
# Spring overview

- Strong concentration gradient
- Low concentrations west of RMNP
- High concentrations east of RMNP
- Ammonia peaks in NE Colorado



# RMNP particle timelines

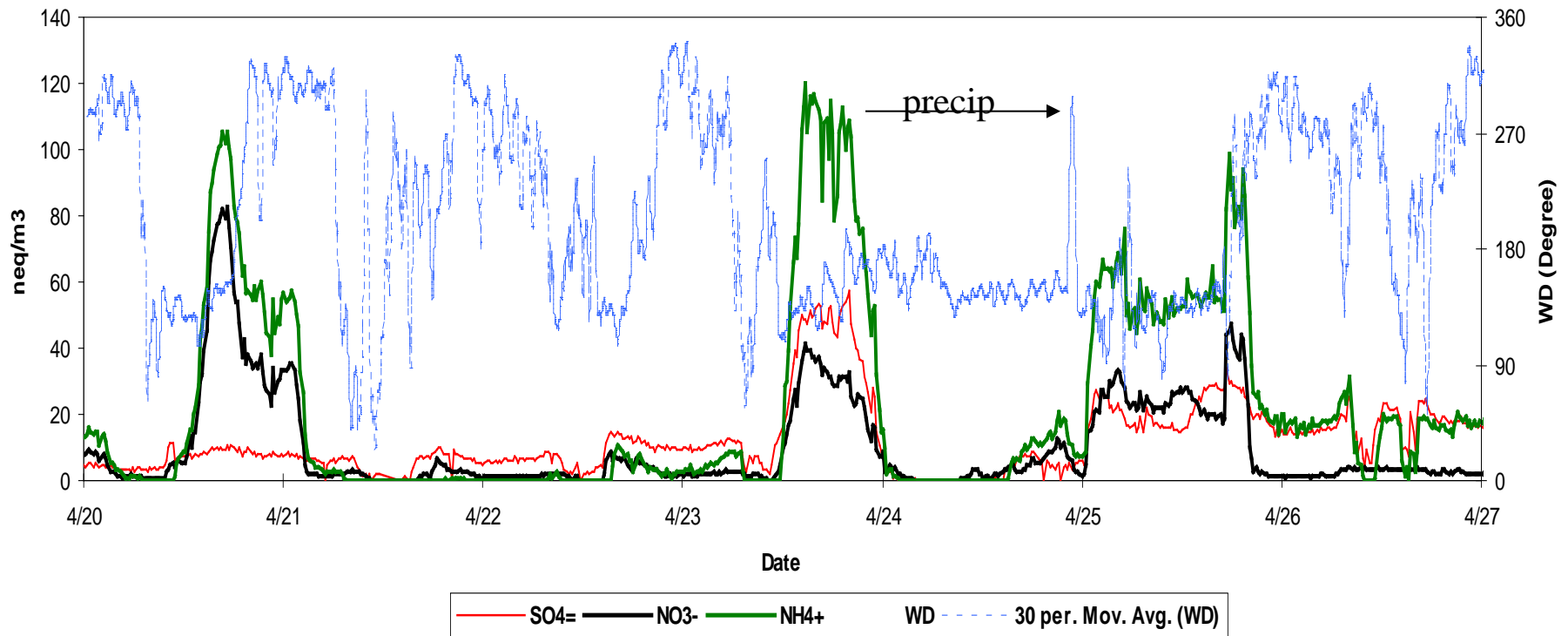
- Highly variable concentrations
- Pollutant mix (sulfate, nitrate, and ammonium) varies between episodes





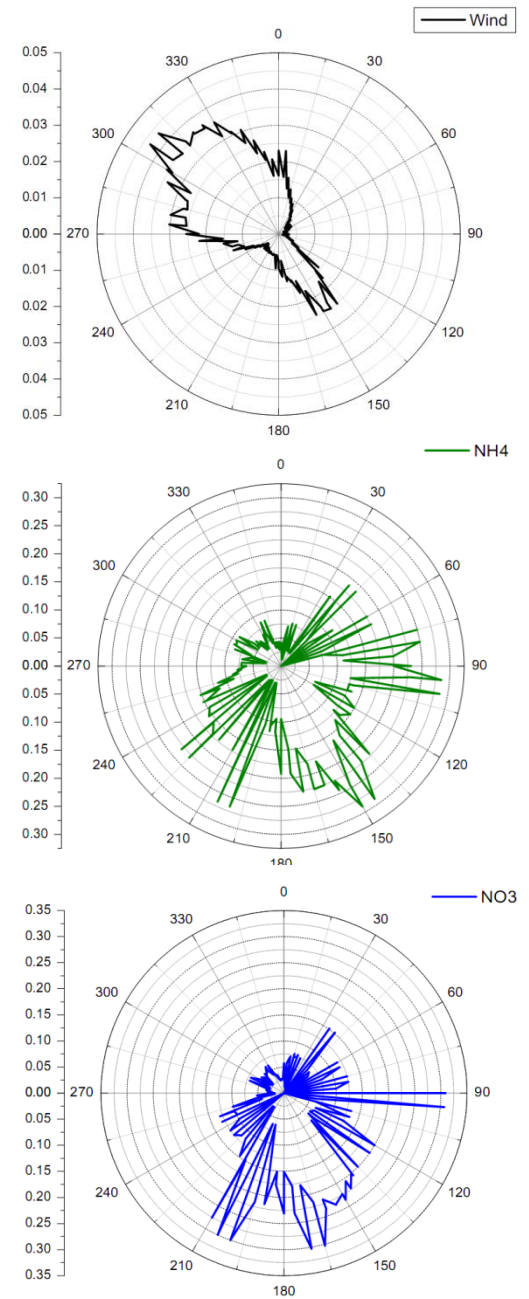
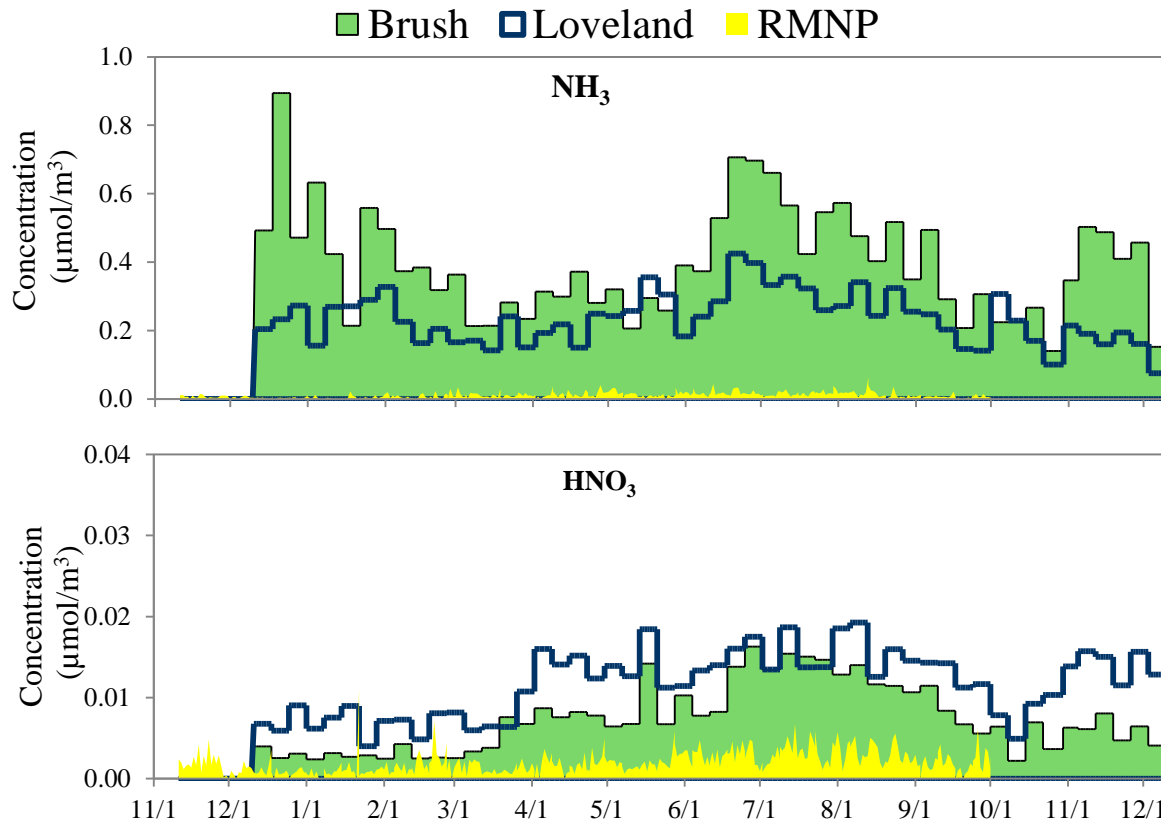
# A closer look...

- Ammonium nitrate episodes associated with upslope flow from east of RMNP



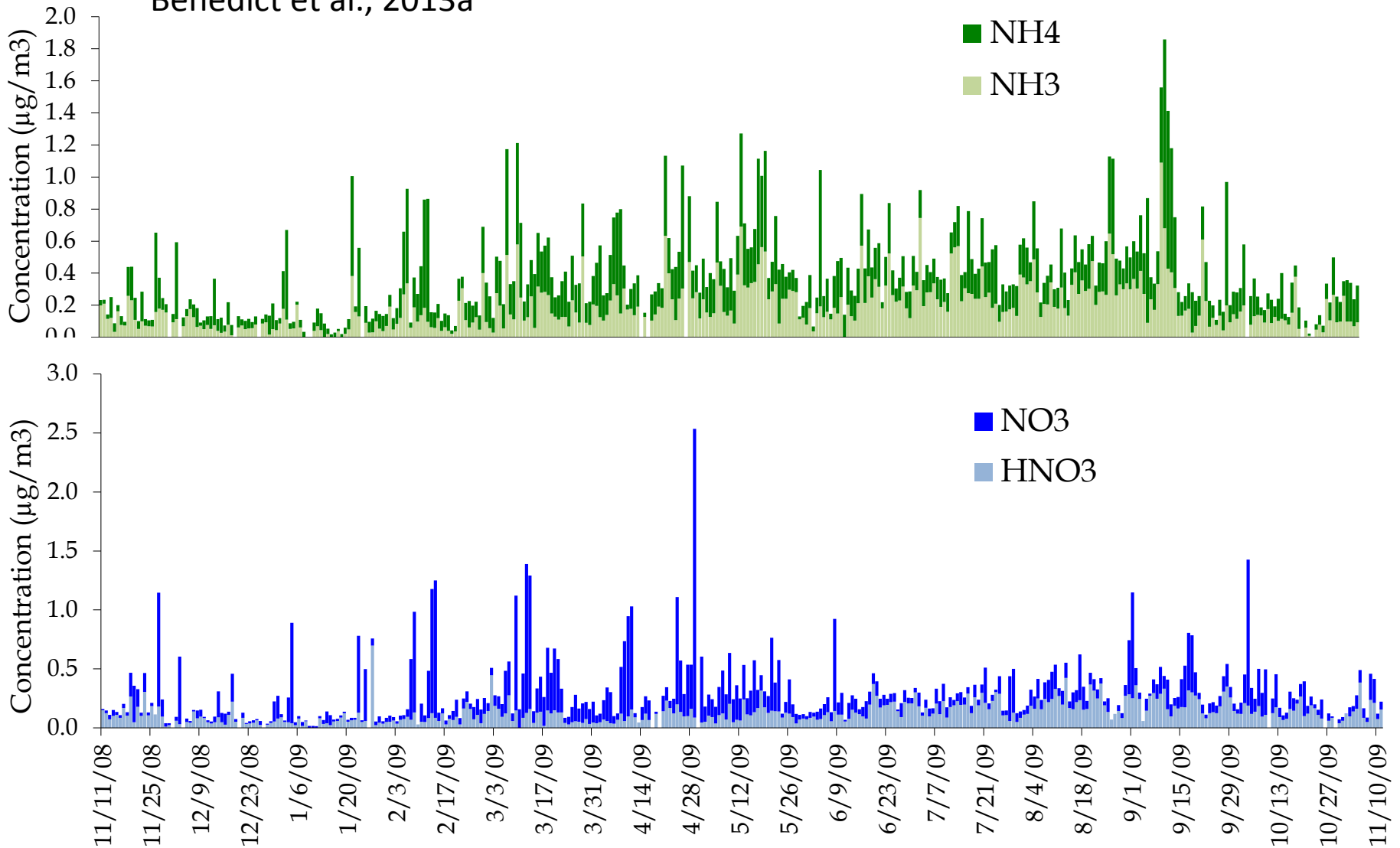


# Annual mountain to plains gradient

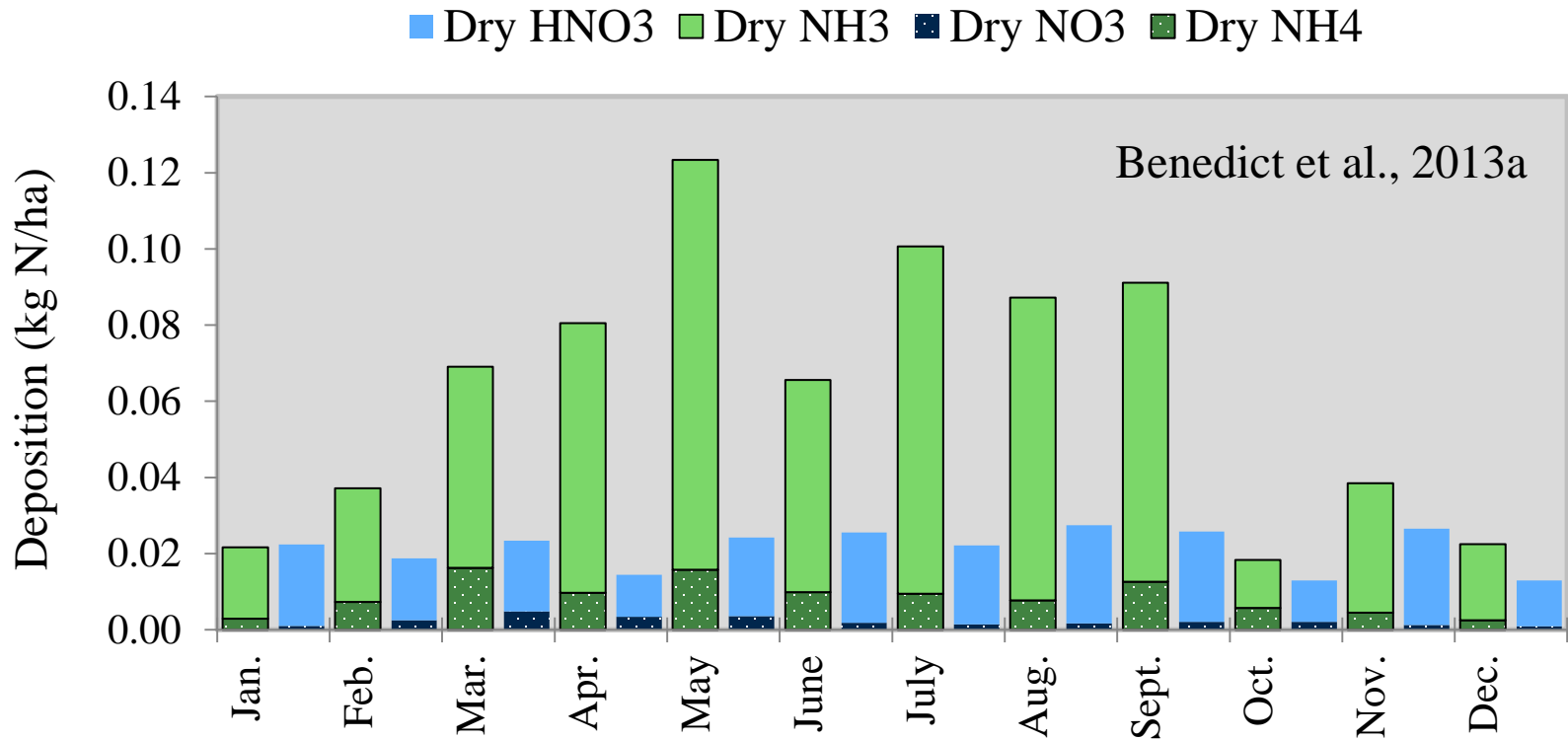


# RMNP reactive N concentrations

Benedict et al., 2013a



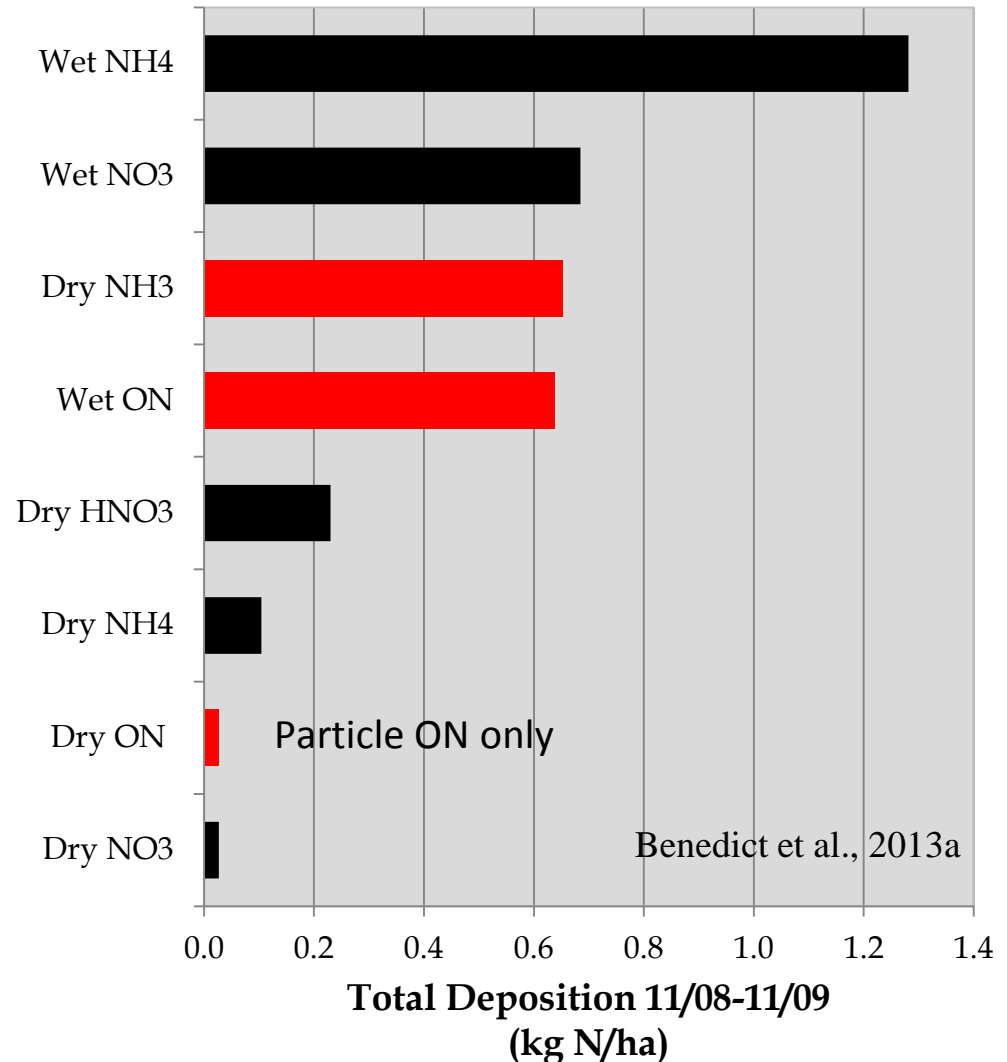
# RMNP seasonal dry deposition budget



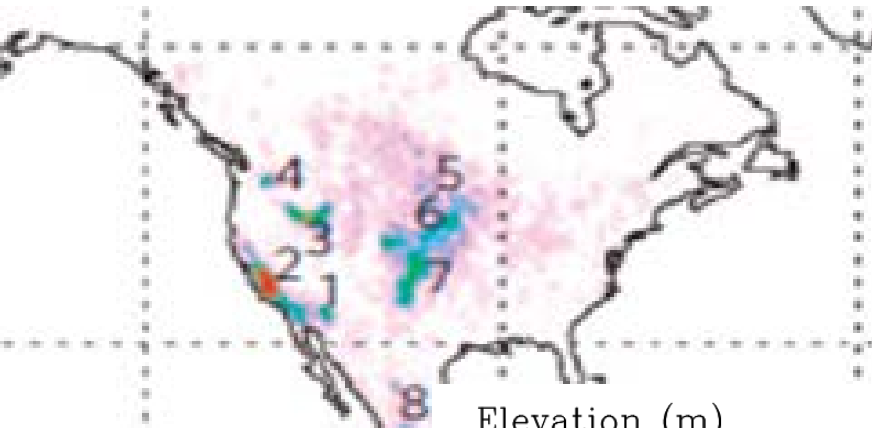
- Ammonia deposition most important; spring and summer peaks

# RMNP N deposition – annual budget

- Wet deposition biggest contributor to N deposition
- Dry deposition strongly dominated by  $\text{NH}_3$

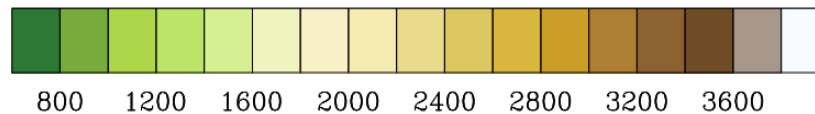
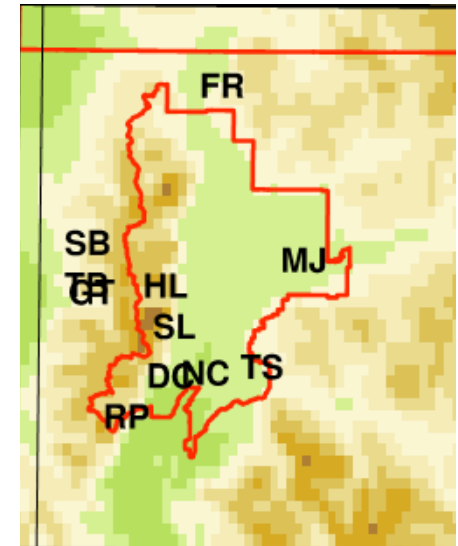
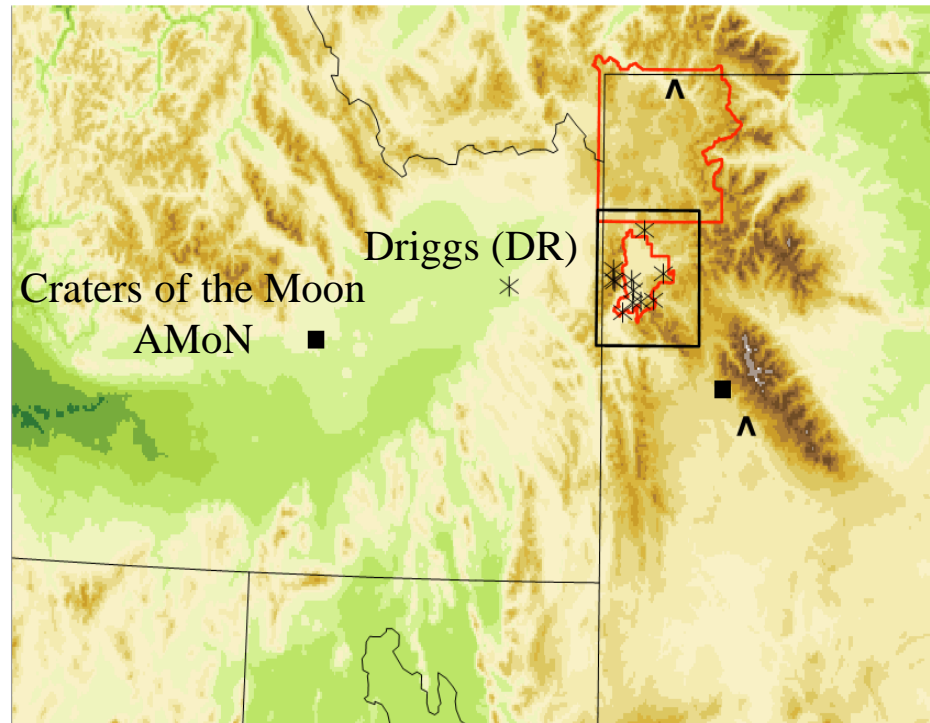


# The 2011 Grand Teton Reactive Nitrogen Deposition Study (GrandTReNDS)



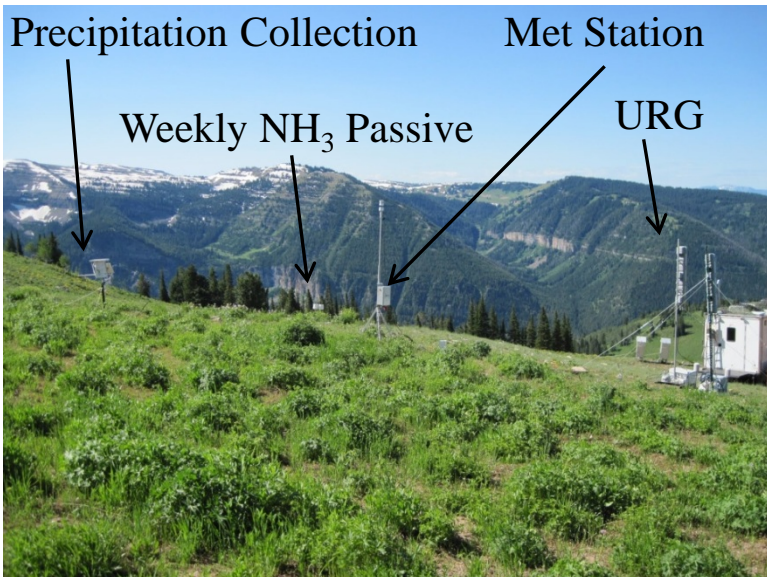
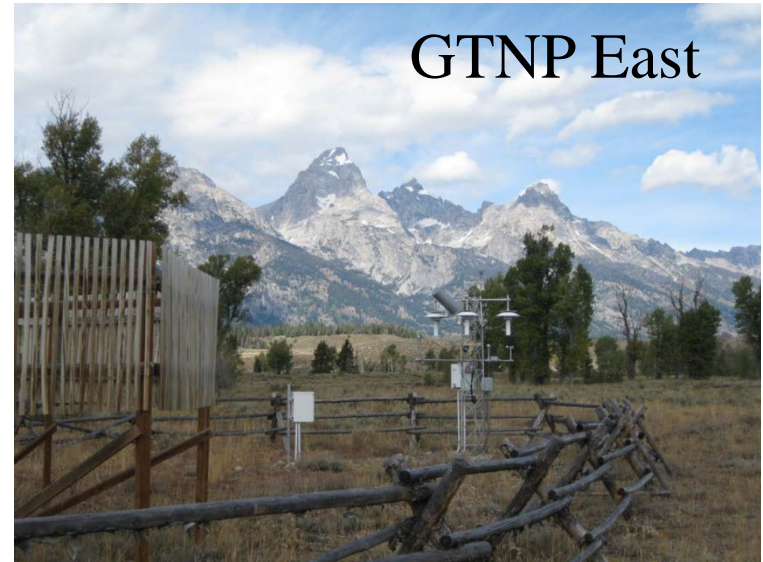
Elevation (m)

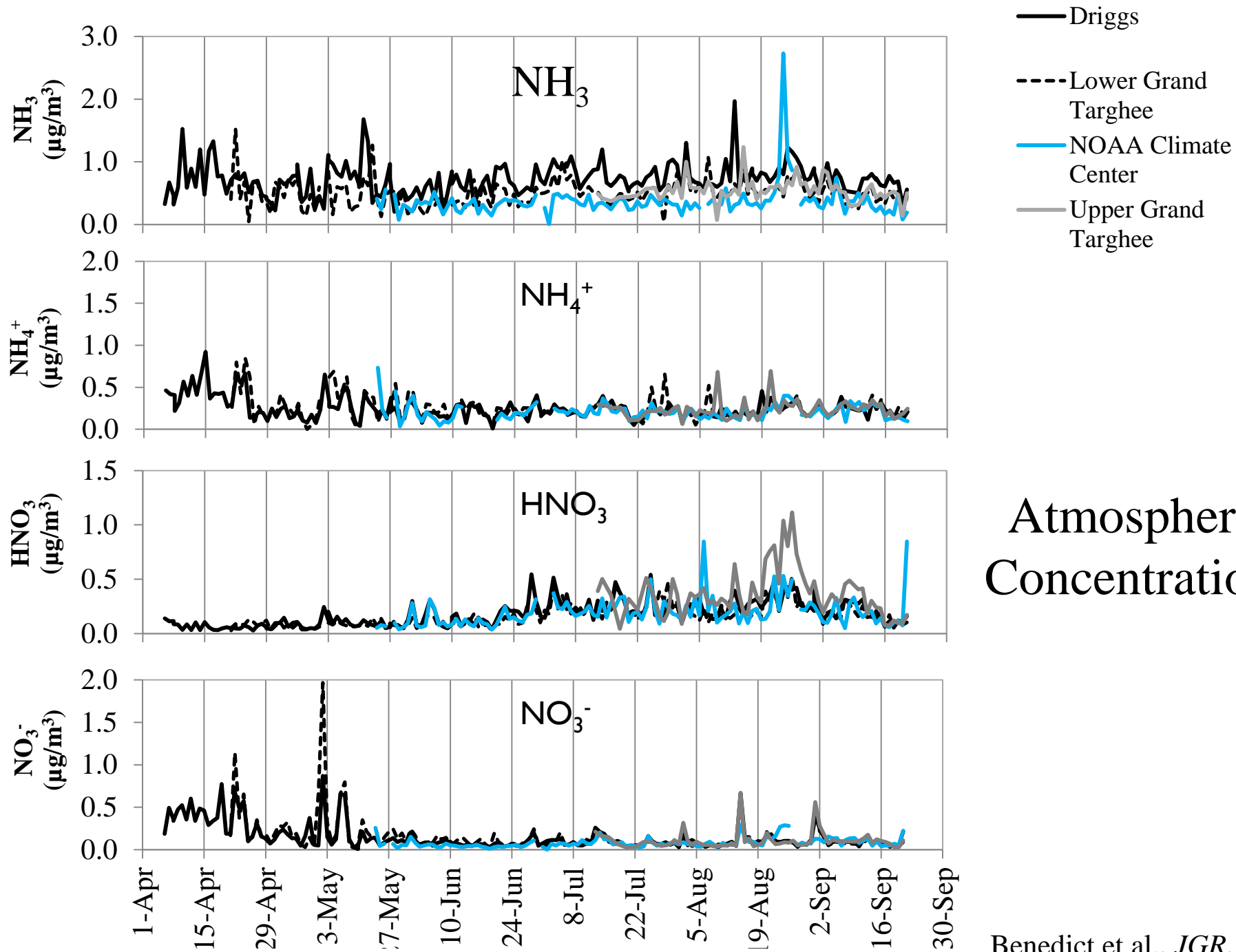
Clarisse et al. (2009)  
IASI satellite  $\text{NH}_3$





# Selected GrandTReNDS sites

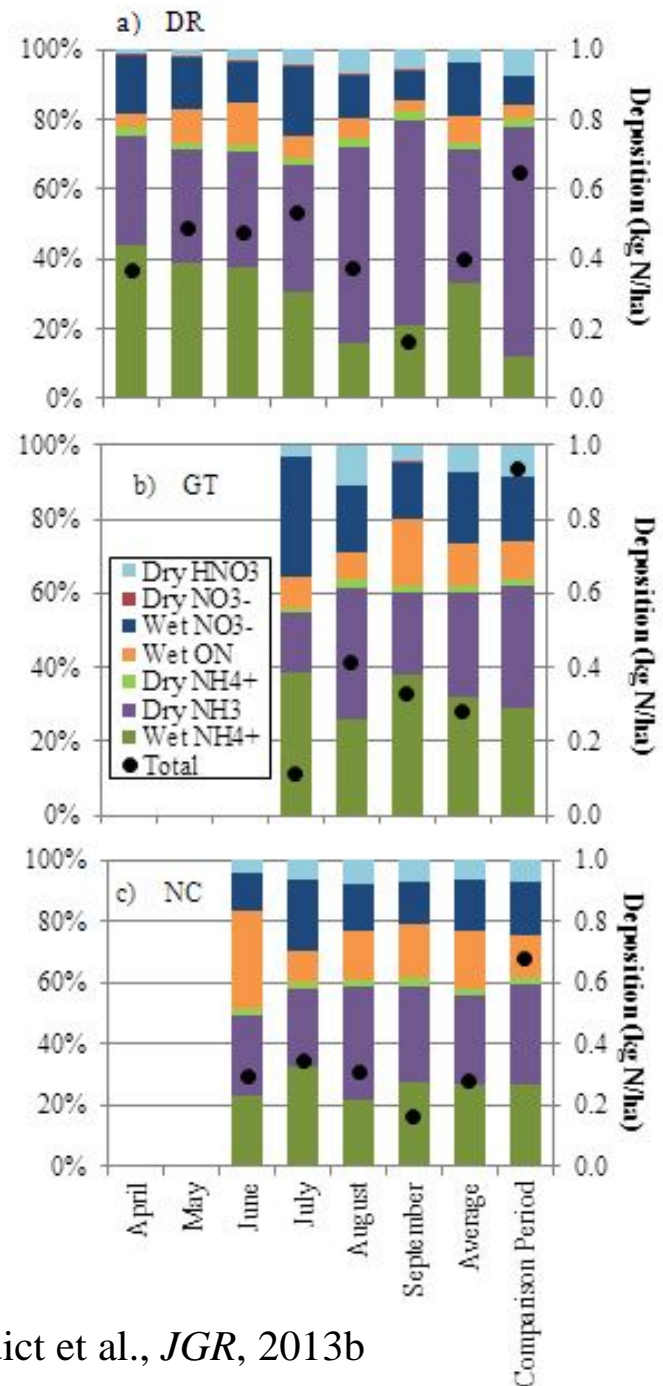




# Atmospheric Concentrations

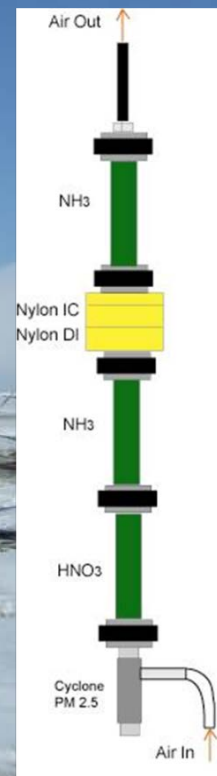
# GrandTReNDS deposition budgets

- Reduced nitrogen comprises 50-80% of the N deposition budget
  - even more important here than in Rocky Mountain National Park



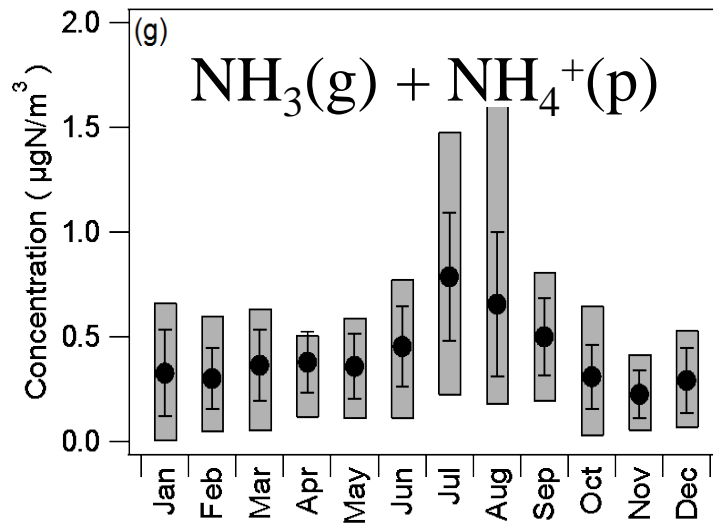


# Long-term measurements of the $\text{NH}_x$ - $\text{NO}_x$ - $\text{SO}_x$ system in Boulder, WY – one of the largest U.S. natural gas producing regions

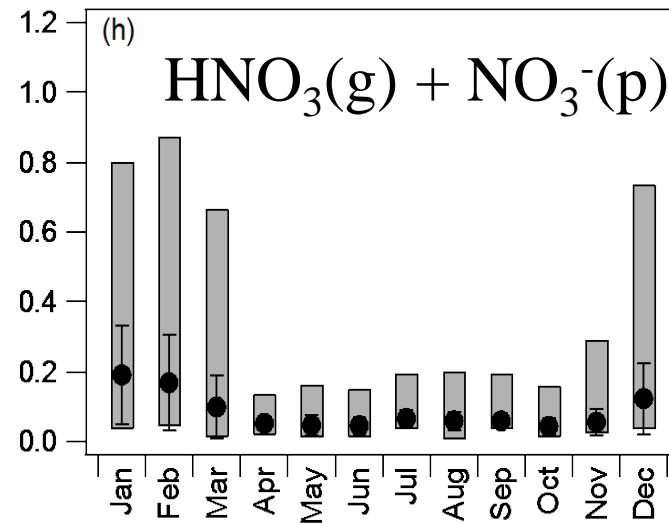


CSU, Air Resource Specialists, Shell

How much  $\text{NH}_3$  is available to react with  $\text{NO}_x$  oxidation products to generate fine particles and lead to haze formation?

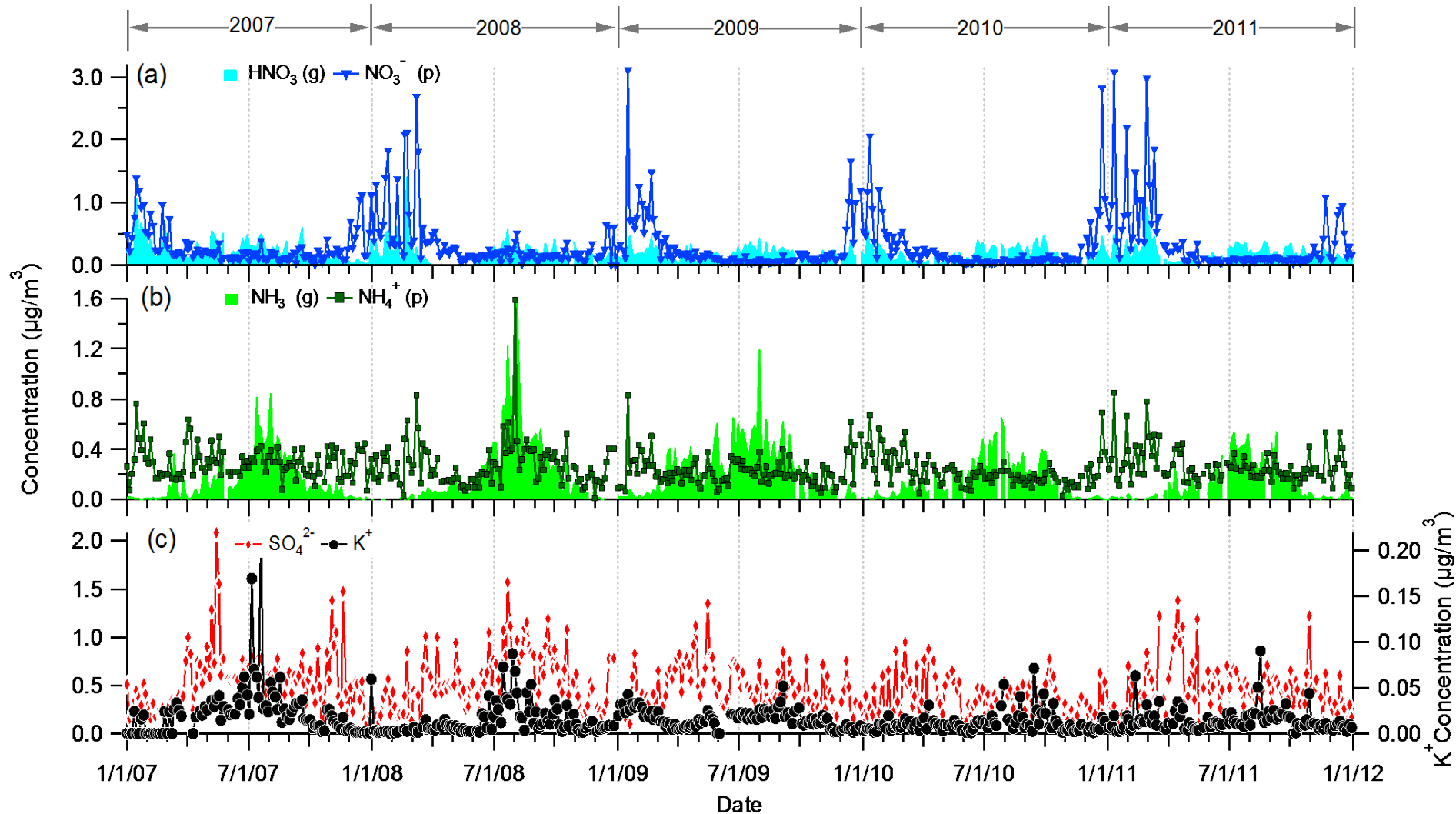


Reduced nitrogen shows typical summer max



Oxidized nitrogen shows unusual winter max tied to winter photochemical smog



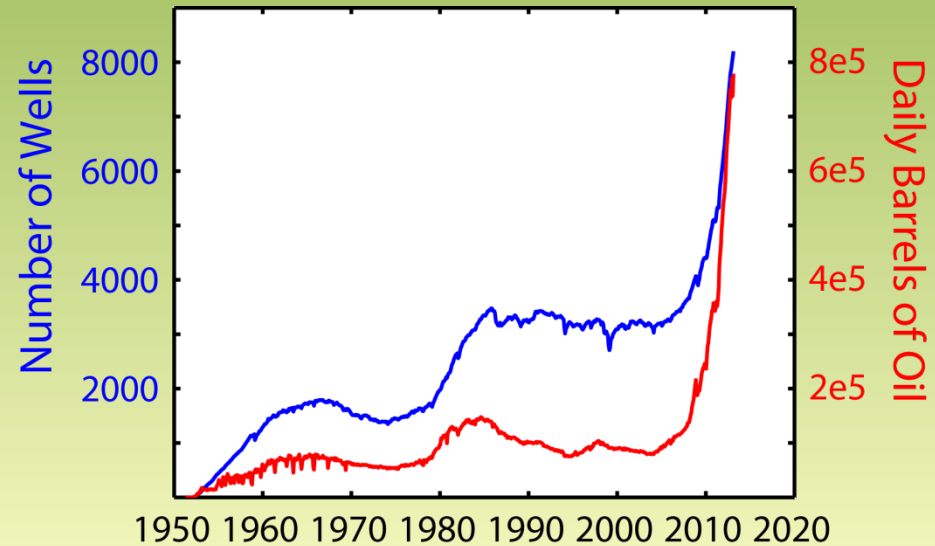


- Active winter photochemistry produces high ozone and nitric acid
- Winter fine particle nitrate formation limited by ammonia availability

# Bakken Shale Oil and Gas Development



## Oil Production in North Dakota

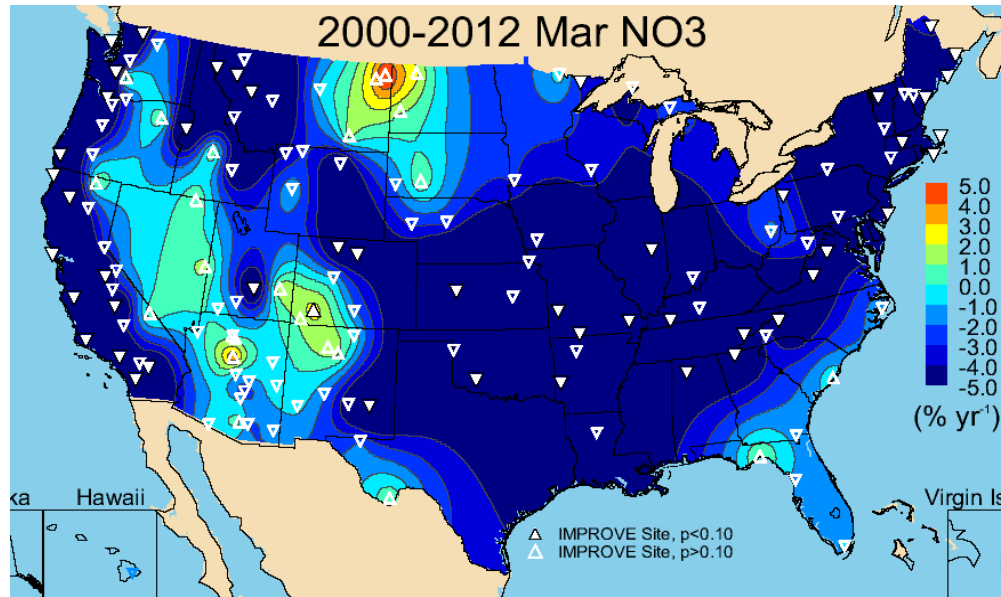


<https://www.dmr.nd.gov/oilgas/>

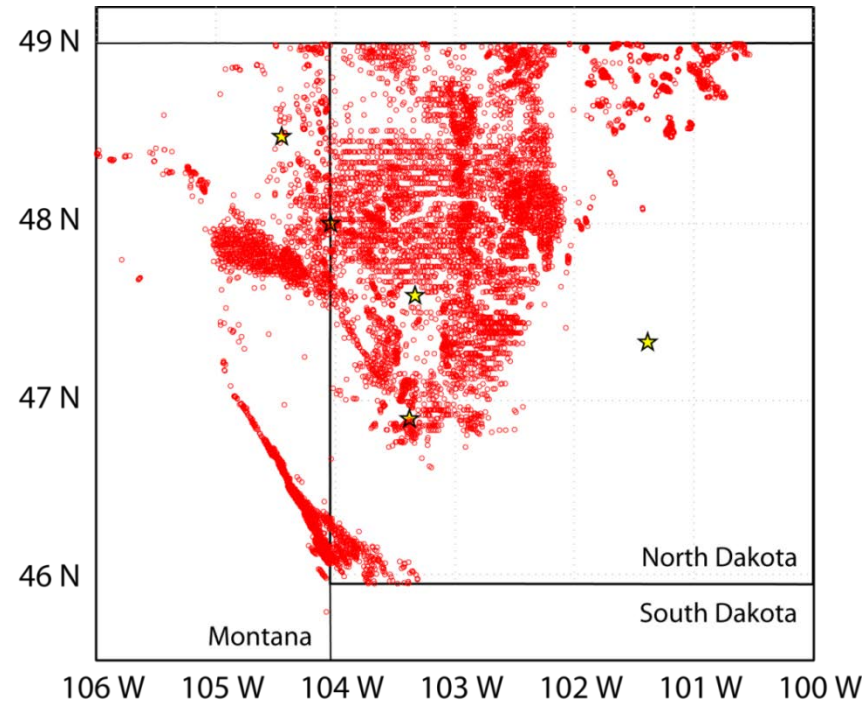


Photo: W. Malm

# Air quality in the Bakken shale region



From J. Hand

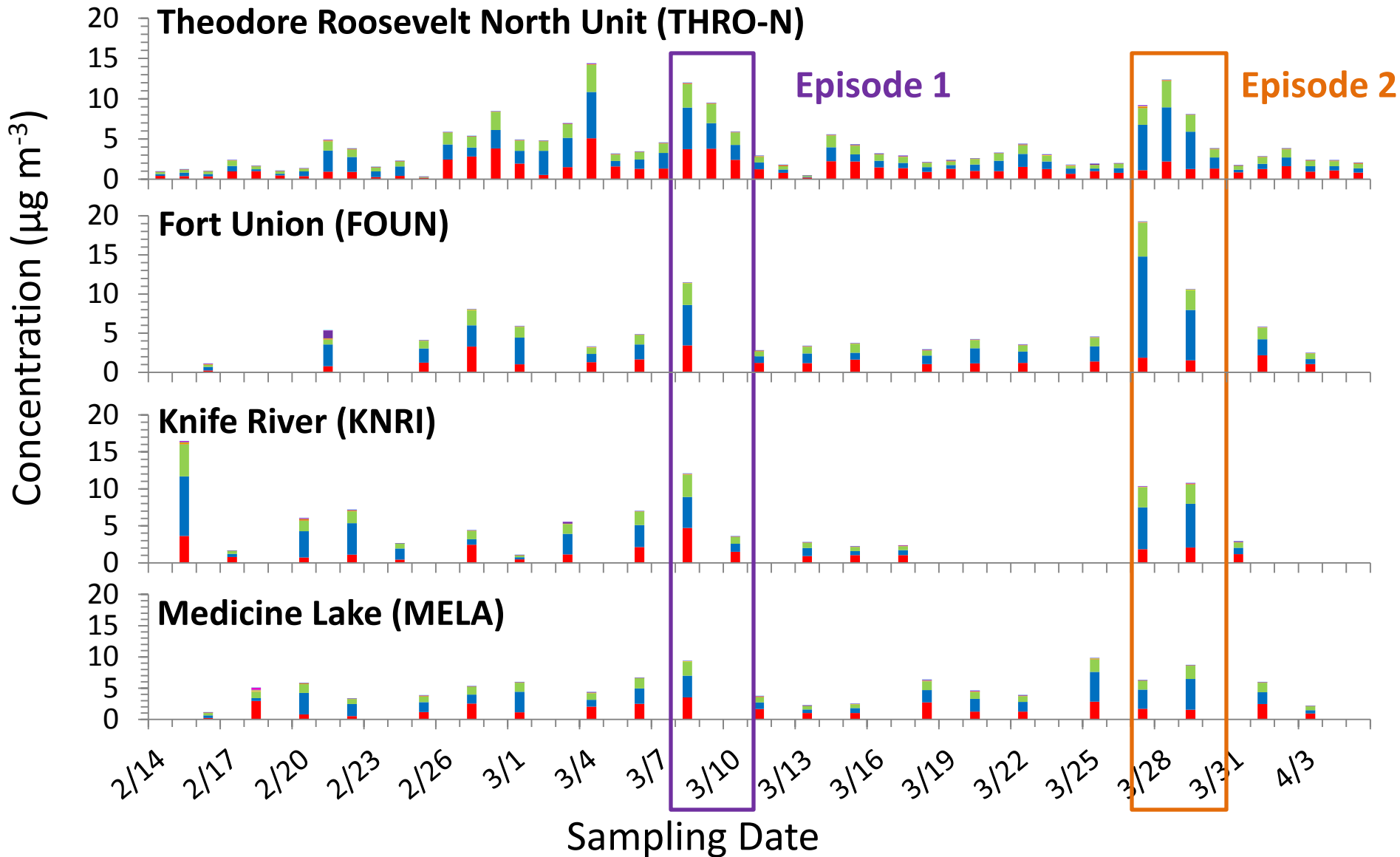


Winter fine particle nitrate increasing in parts of western U.S.

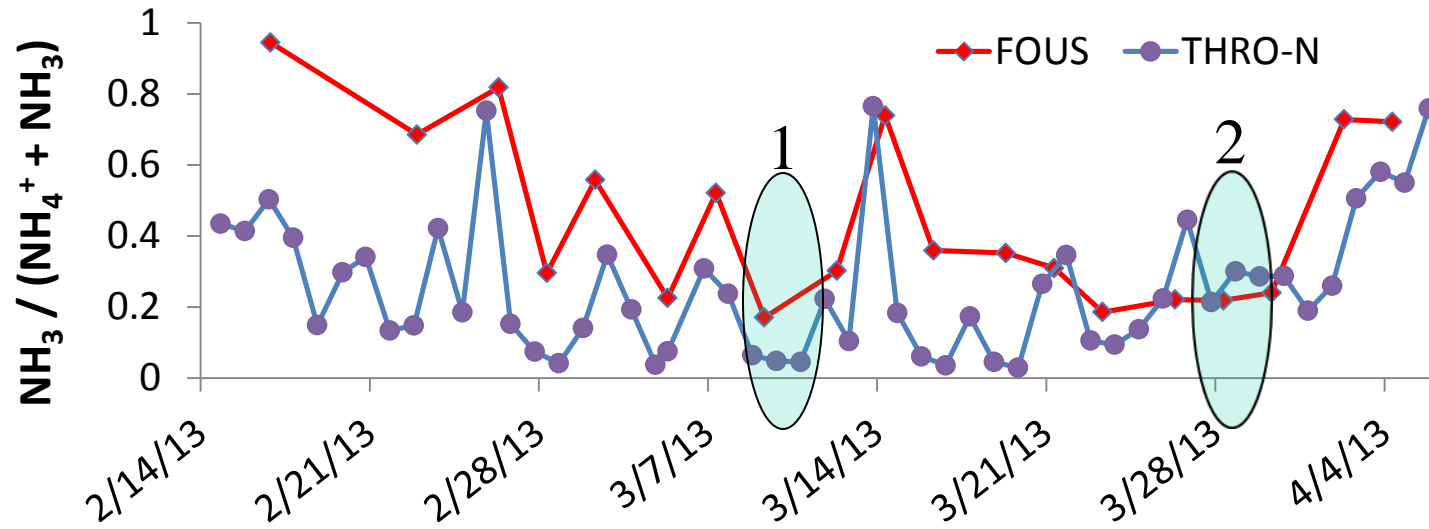
Winter 2013 study conducted across region of extensive oil well drilling to examine air quality impacts

# Bakken Particle Composition

Legend: Nitrite (light blue), Chloride (magenta), Sodium (purple), Potassium (yellow), Magnesium (brown), Calcium (orange), Ammonium (light green), Nitrate (dark blue), Sulfate (red)



# Does ammonia limit $\text{NH}_4\text{NO}_3$ formation in Bakken?



- Ammonia nearly depleted in Theodore Roosevelt National Park during episode 1, but not episode 2.
- Excess ammonia always available at Fort Union.
  - $\text{HNO}_3$  currently limits  $\text{NH}_4\text{NO}_3$  formation

**HOW CAN WE ROUTINELY  
MEASURE  $\text{NH}_3$  AND  $\text{NH}_4^+$  ?**

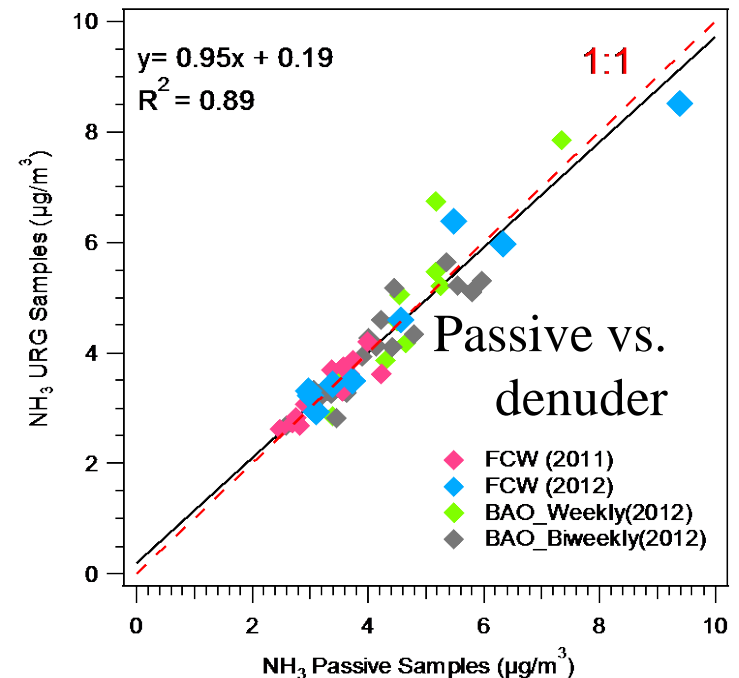
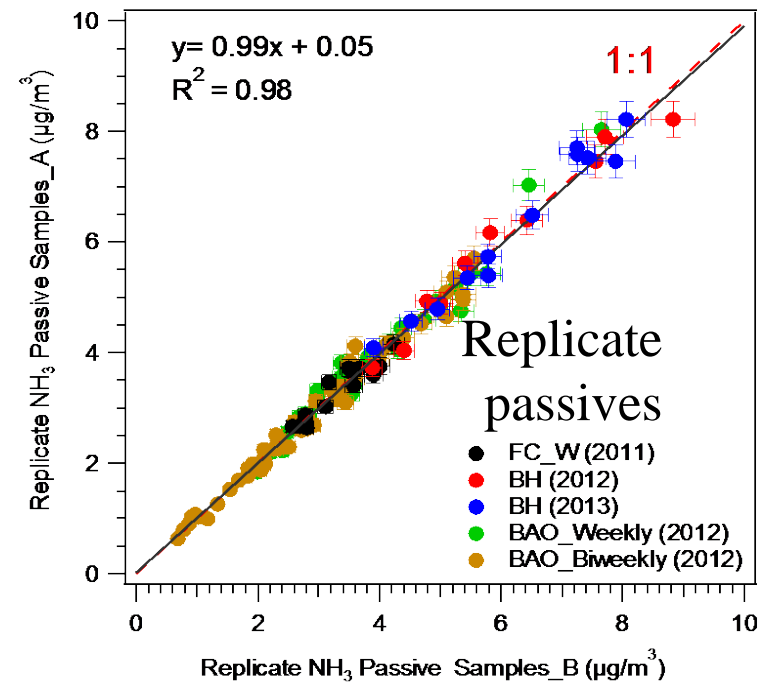
# U.S. network measurements of inorganic N species concentrations

- $\text{NO}_3^-$ 
  - CASTNet quantifies  $\text{PM}_{2.5}$  deposition
  - $\text{PM}_{2.5}$  concentrations also measured by IMPROVE and CSN
- $\text{HNO}_3$ 
  - CASTNet quantifies weekly concentration/deposition
- $\text{NH}_4^+$ 
  - CASTNet quantifies weekly concentration/deposition
- $\text{NH}_3$ 
  - AMoN (bi-weekly)
  - Regional networks (e.g., SEARCH)
  - Short-term/special studies

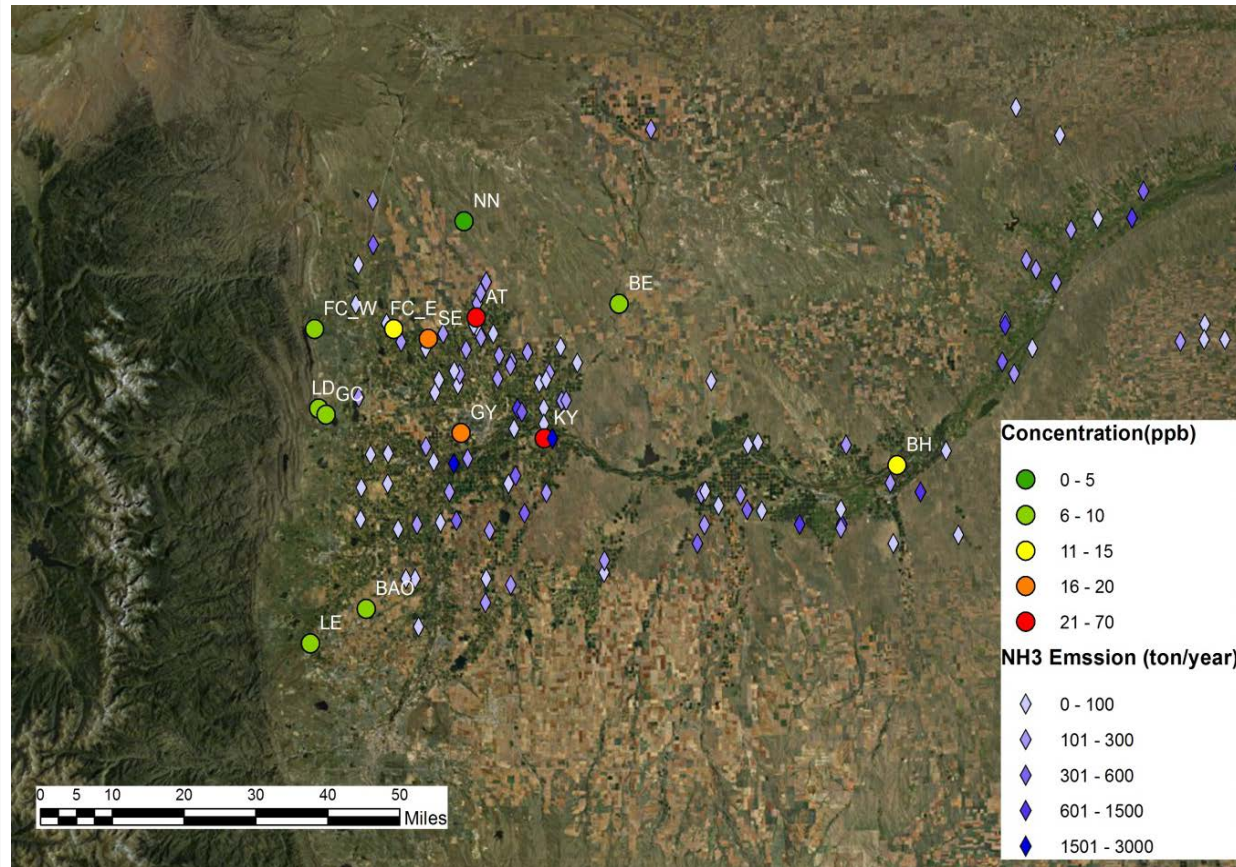
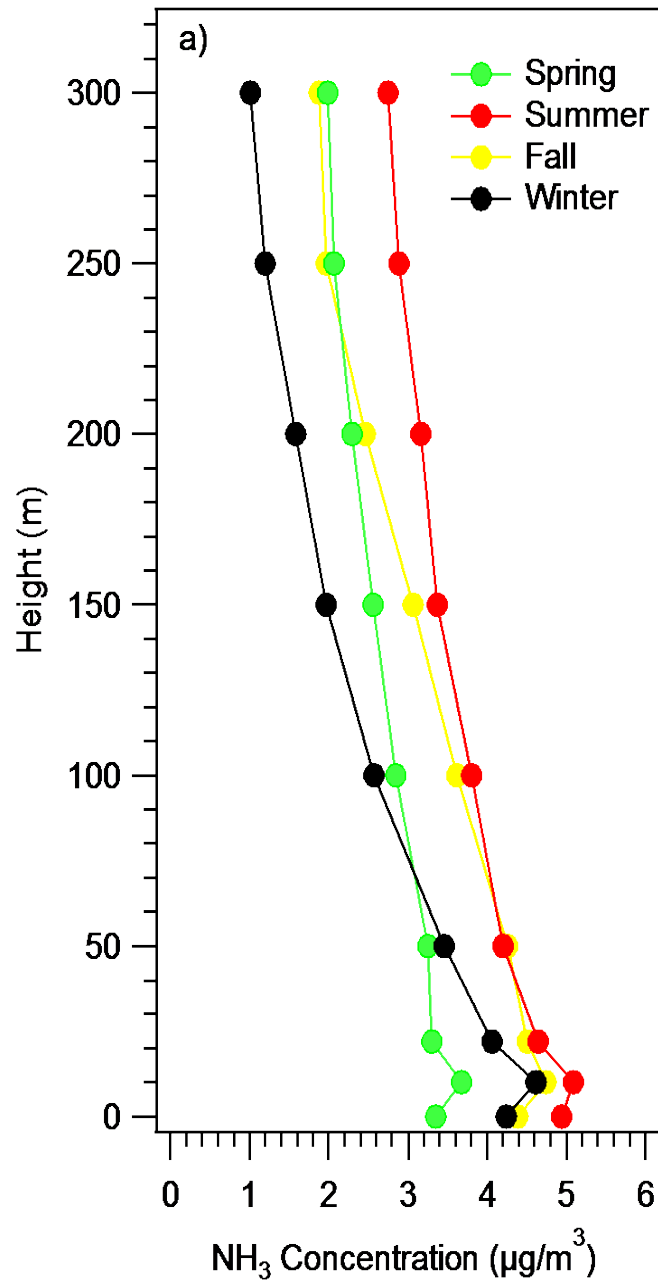


# Radiello passive NH<sub>3</sub> samplers

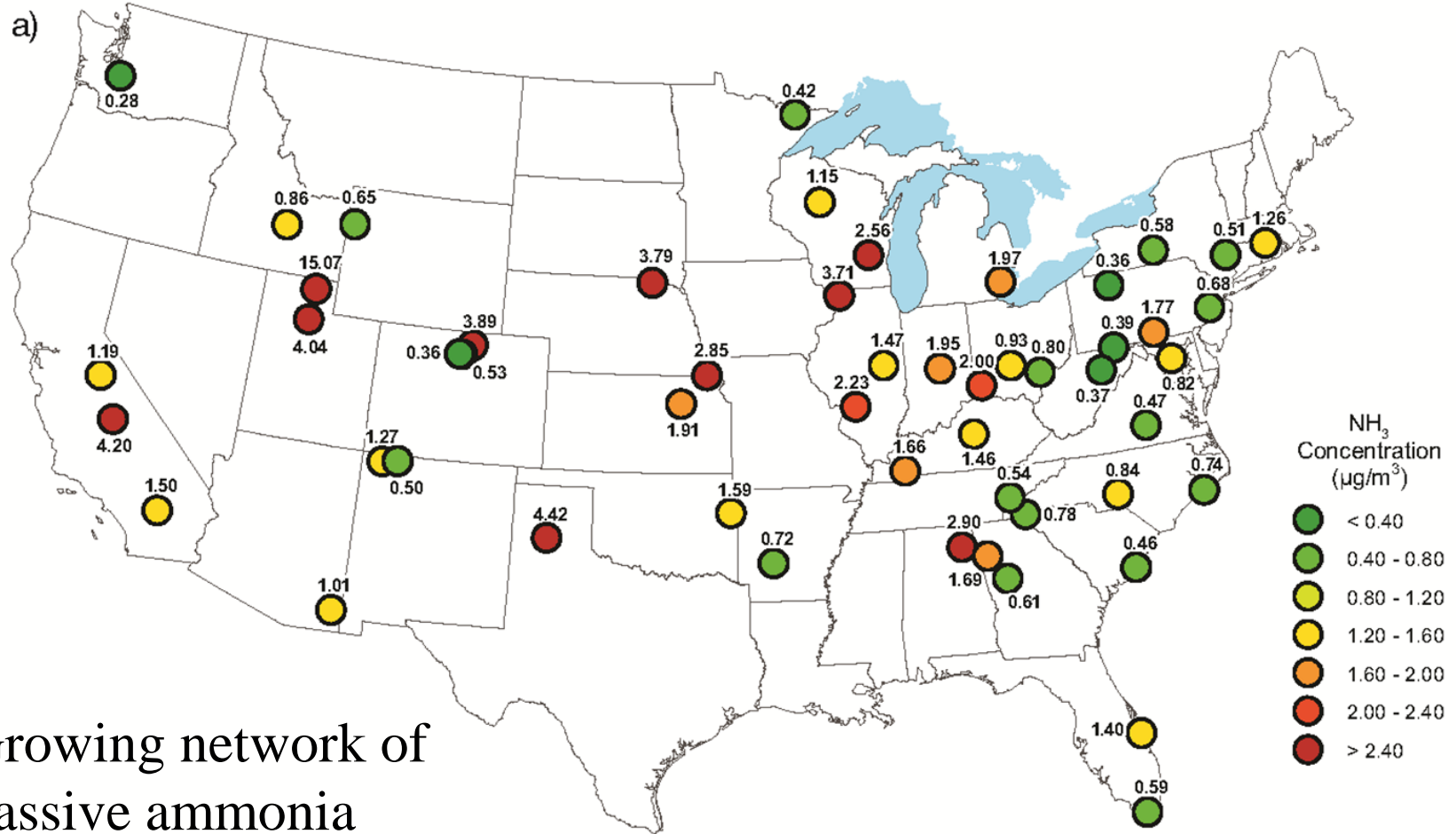
- Inexpensive
- Greater spatial coverage
- ~1-2 week time resolution
- Excellent precision and good accuracy
- NH<sub>3</sub> only



# NE Colorado spatial NH<sub>3</sub> gradients



# NADP AMoN Network



- Growing network of passive ammonia sampling
- Ideal for  $\text{NH}_3$  deposition

Figure courtesy of Chris Lehmann

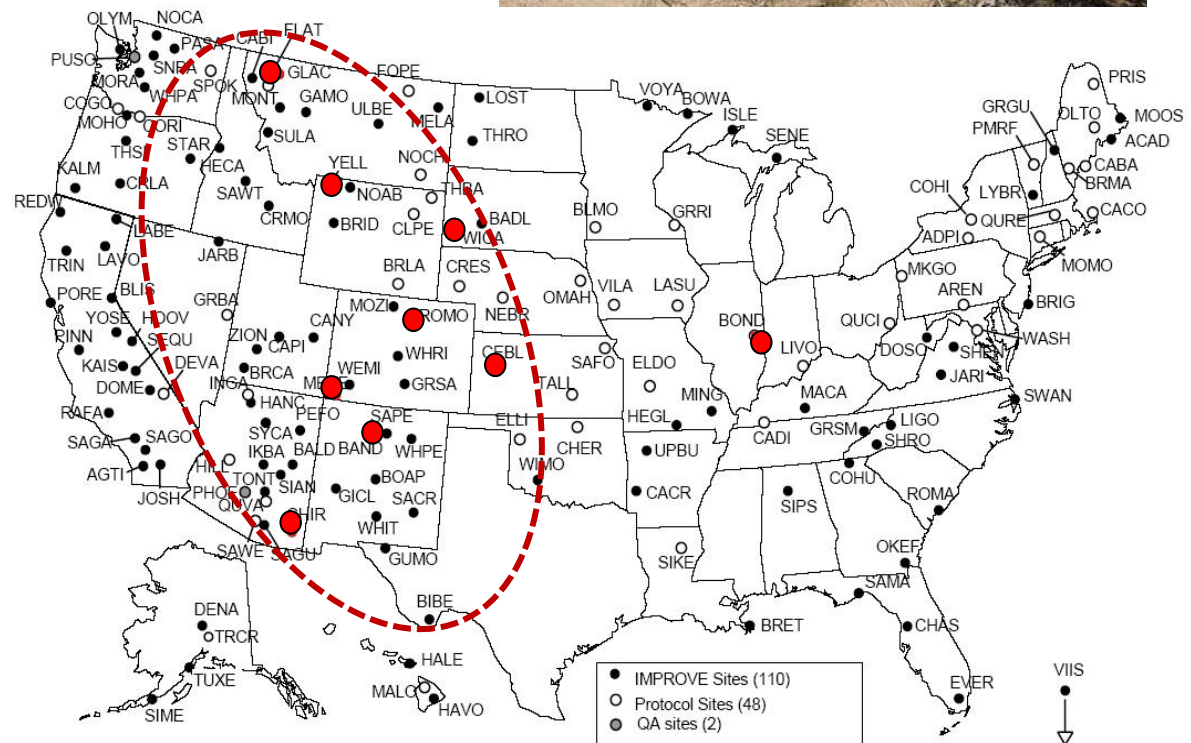
## Limitations to AMoN passive sampling approach

- Low time resolution (1-2 weeks is typical) makes source attribution difficult
- Does not measure  $\text{NH}_4^+$ 
  - Leaves total  $\text{NH}_x$  burden unconstrained
  - Limits ability to understand PM formation
  - Limits ability to validate model simulations

# Pilot IMPROVE NH<sub>x</sub> network



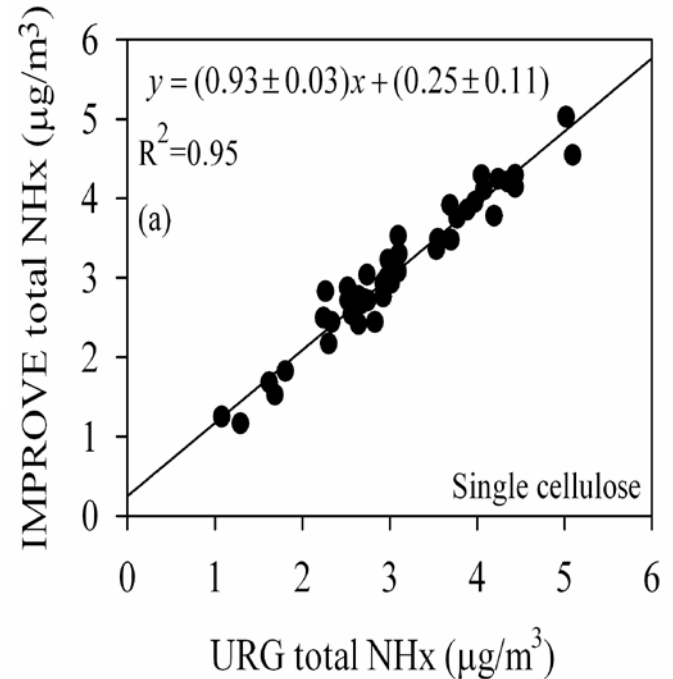
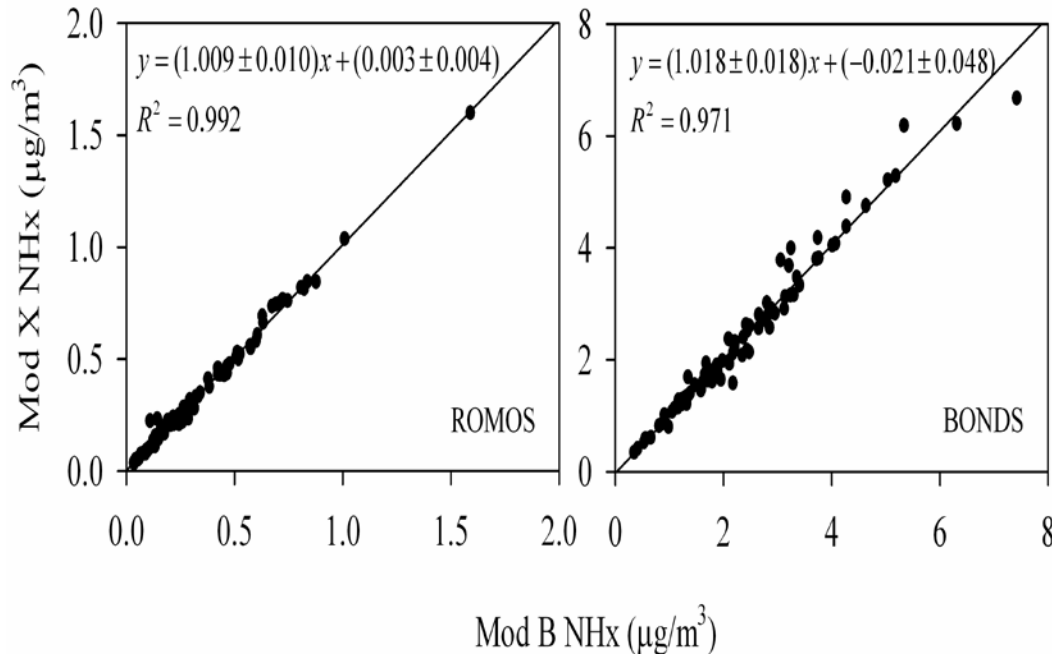
- Rocky Mountain focus
  - 9 sites, 1-in-3 day sampling
  - 4/2011 - 8/2012
- Single phosphorous acid-coated filter to capture  $\text{NH}_4^+$  +  $\text{NH}_3$





# NH<sub>x</sub> measurement quality

- Good accuracy vs. reference URG denuder/filter-pack (denuder + filter + backup denuder)

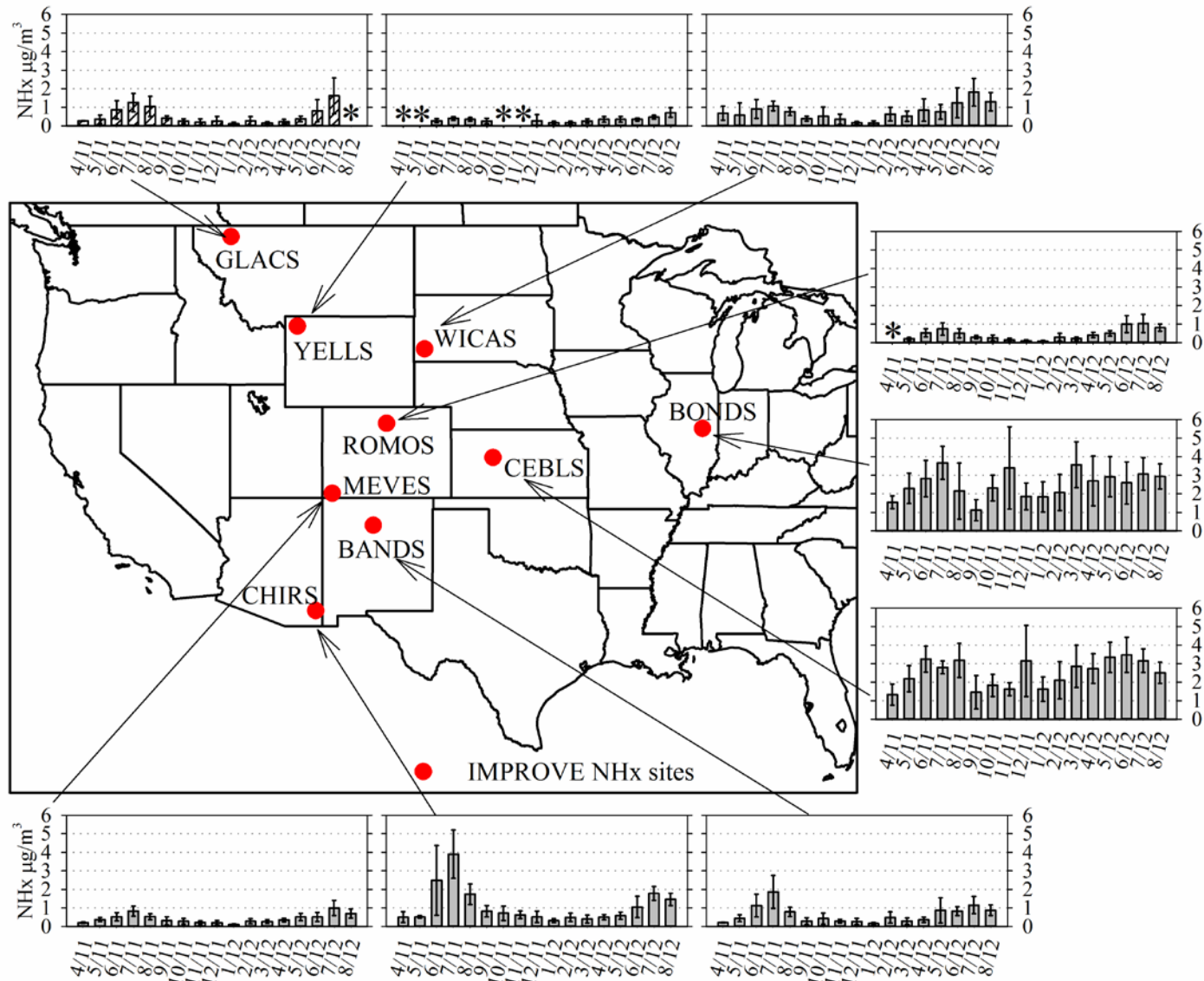


- Excellent field precision demonstrated by co-located samplers

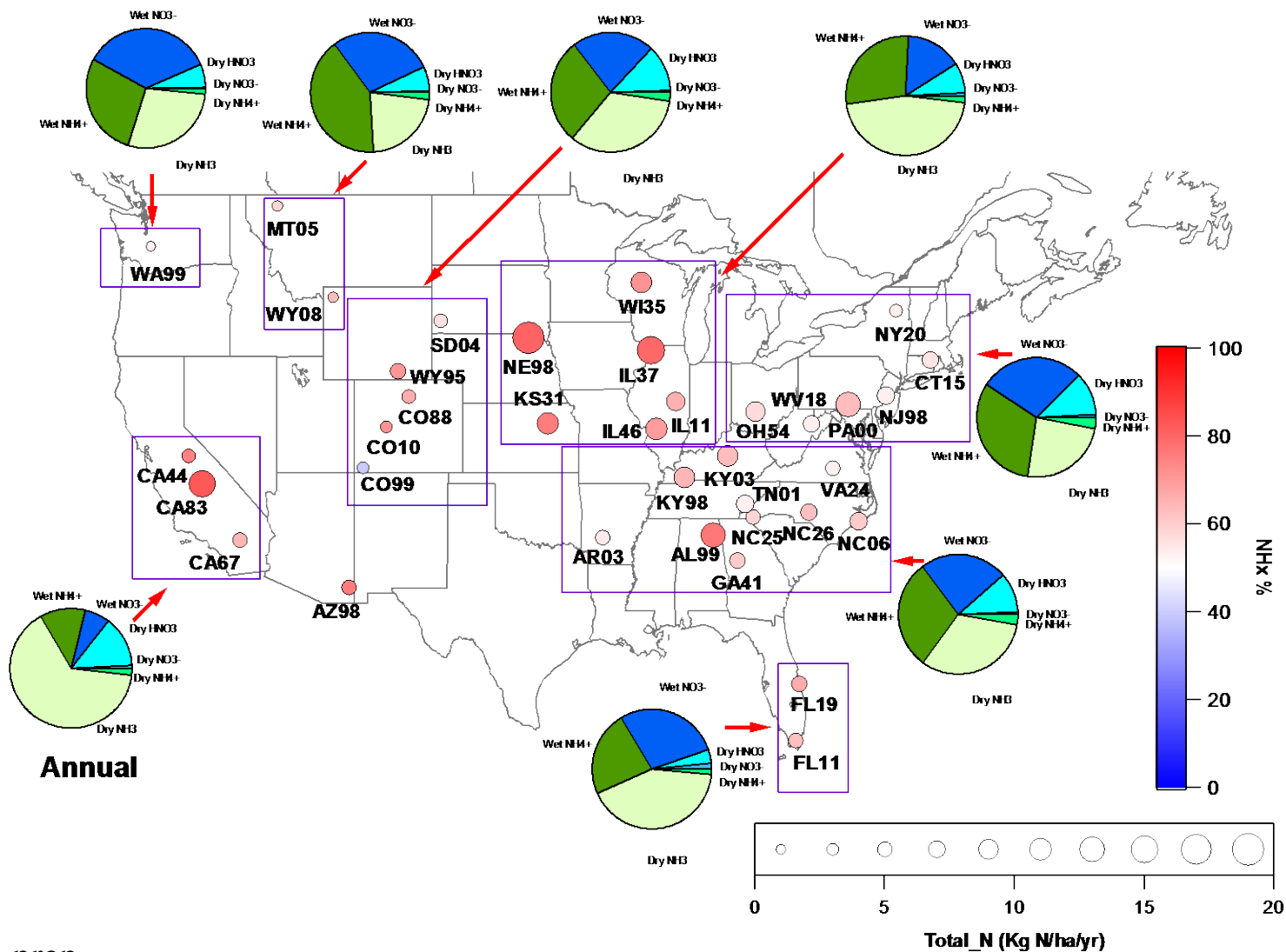


# IMPROVE NH<sub>x</sub> data overview

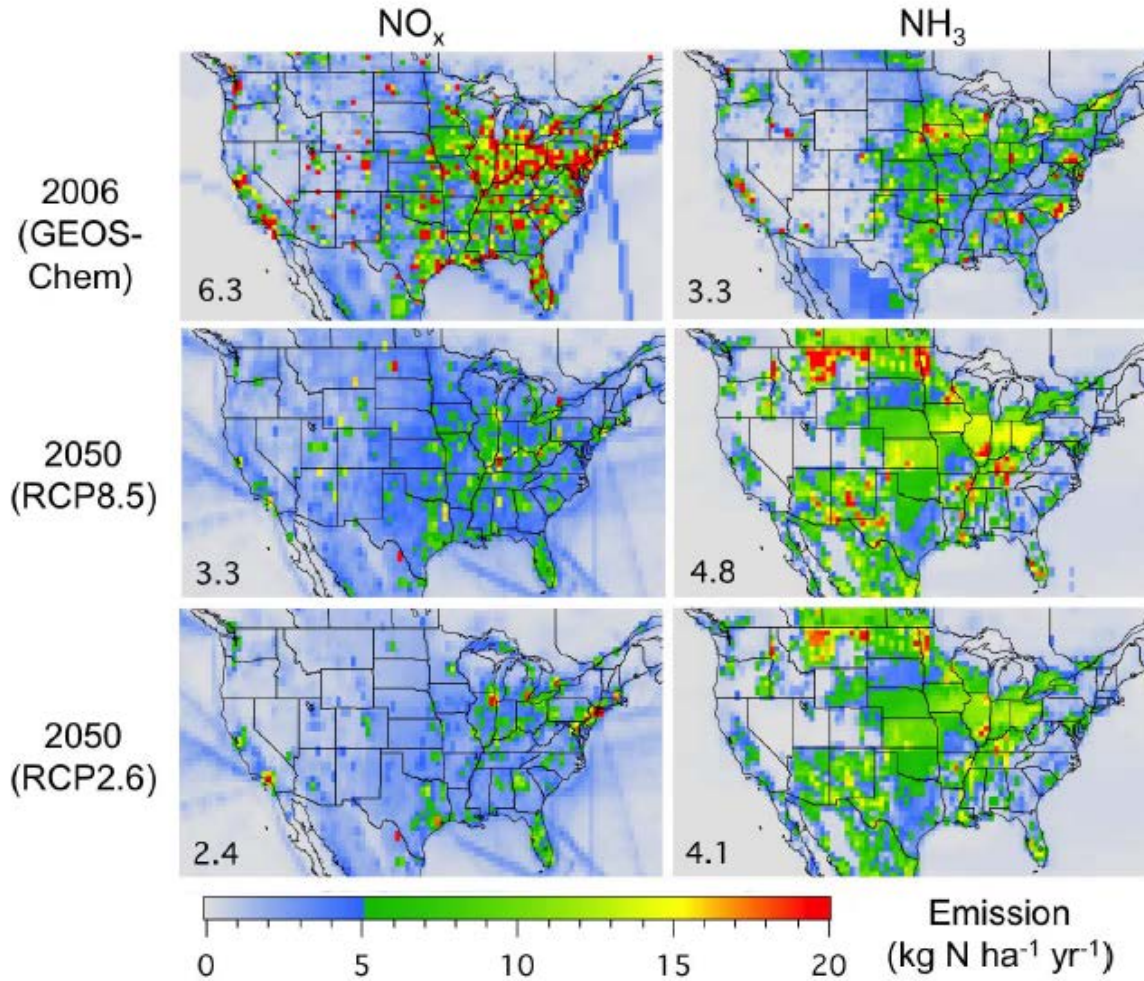
- Highest concentrations at eastern and southern sites
- Seasonal cycle apparent at all sites
- Fall/winter secondary max in ag regions
- Fire influence at Chiricahua in 2011



# Total N deposition budgets: wet + dry



# Anticipated future emissions changes will make $\text{NH}_x$ increasingly even more important



$\text{NO}_x$  expected to decrease while  $\text{NH}_3$  increases

Ellis et al., 2012

**Fig. 1.**  $\text{NO}_x$  and  $\text{NH}_3$  emissions in North America for 2006 and 2050. Numbers inset give contiguous US totals ( $\text{Tg N yr}^{-1}$ ). 2050 emissions are from the RCP8.5 and RCP2.6 scenarios

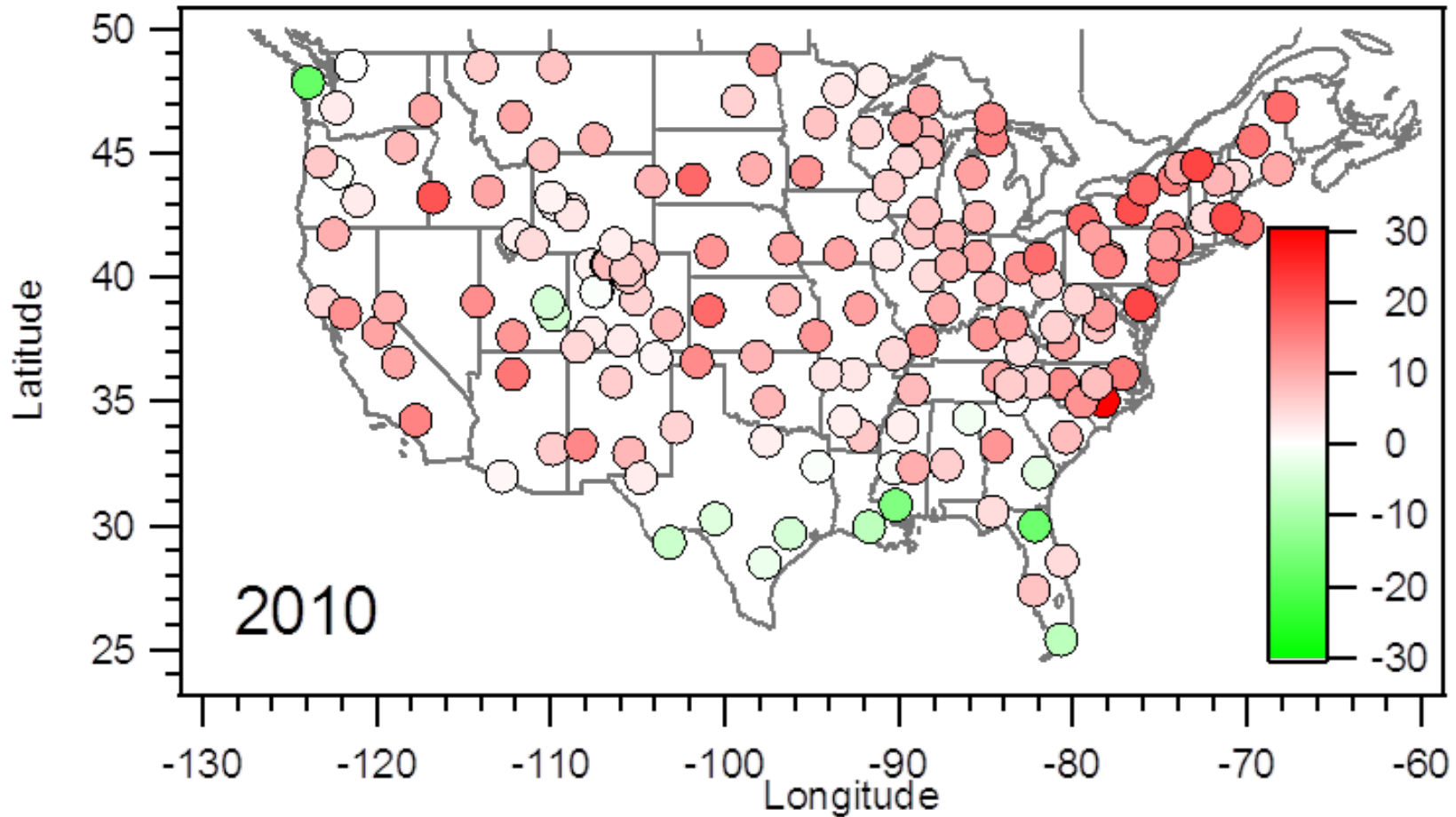
# Summary

- Ammonia is an important contributor to dry & wet deposition of reactive N
- As  $\text{NO}_x$  emissions decrease,  $\text{NH}_x$  is an increasingly important component of N deposition
- Ammonia also a key factor affecting PM acidity and  $\text{NH}_4\text{NO}_3$  formation
  - Abundant ammonia supports greater  $\text{NH}_4\text{NO}_3$  PM formation and haze
- Network measurements needed to routinely characterize  $\text{NH}_3$  and  $\text{NH}_4^+$  at higher time resolution to
  - Understand PM formation
  - Validate air quality models
  - Attribute source contributions





# Increase in $\text{NH}_4^+$ fraction



- A sizable increase in the  $\text{NH}_4^+$  fraction (absolute % change) of wet inorganic nitrogen deposition is seen across most of the country (2010 – 1989)